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Assistant Commissioner for Patents Box Patent Applications Washington D.C.

Attorney Docket No.67242/107 (must include alphanumeric codes if no inventors named)

UTILITY PATENT APPLICATION TRANSMITTAL (new nonprovisional applications under 37 CFR 1.53(b))

Transmitted herewith for filing is the patent application of:

Fumihiko WATANABE, Hiroshige TSUZUKI, and Mitsuaki OHTANI **INVENTORS:**

SULFONATED AMINO ACID DERIVATIVES AND METALLOPROTEINASE TITLE: INHIBITORS CONTAINING THE SAME

In connection with this application, the following are enclosed:

APPLICATION ELEMENTS:

<u>XX</u> Specification - <u>113</u> TOTAL PAGES Ū (preferred arrangement:) لِيًا -Descriptive Title of the Invention -Cross Reference to Related Applications -Statement Regard Fed sponsored R&D -Reference to Microfiche Appendix ÷.,, -Background of the Invention -Brief Summary of the Invention
-Brief Description of the Drawings (if filed) IJ -Detailed Description Ü -Claim(s) -Abstract of the Disclosure Drawings - Total Sheets <u>0</u> XX Declaration and Power of Attorney - Total Sheets 2 XX Newly executed (original or copy) _ Copy from a prior application (37 CFR 1.63(d)) (relates to continuation/divisional boxes completed) - NOTE: Box below ___ DELETION OF INVENTOR(S) - Signed statement attached deleting inventor(s) named in the prior application, see 37 CFR 1.63(d)(2) and 1.33(b). XX Incorporation By Reference The entire disclosure of the prior application is considered as being part of the disclosure of the accompanying application and is hereby incorporated by reference therein. Microfiche Computer Program (Appendix) Nucleotide and/or Amino Acid Sequence Submission (if applicable, all necessary) _ Computer Readable Copy Paper Copy (identical to computer copy)

Statement verifying identify of above copies ACCOMPANYING APPLICATION PARTS

- XX Assignment Papers (cover sheet & document(s))
- _ 37 CFR 3.73(b) Statement (when there is an assignee)
- English Translation Document (if applicable)
- XX Information Disclosure Statement (IDS) with PTO-1449. 17 Copies of IDS Citations XX Preliminary Amendment
- XX Return Receipt Postcard (MPEP 503)

Ut lity Patent Application Transmittal Attorney Docket No. 67242/107 - Foley & Lardner Page 2

__ Small Entity Statement(s)

____ Statement file in prior application, status still proper and desired. ___ Certified Copy of Priority Document(s) with Claim of Priority

(if foreign priority is claimed).

XX OTHER: Check for \$2,170.00

If a **CONTINUING APPLICATION**, check appropriate box and supply the requisite information:

XX Continuation of prior application Serial No. PCT/JP97/00126.

XX Amend the specification by inserting before the first line the following sentence: -- This application is a continuation of application Serial No. PCT/JP97/00126, filed January 22, 1997. --

CORRESPONDENCE ADDRESS:

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FEE CALCULATIONS: (Small entity fees indicated in parentheses.)

7.00		<u> </u>		,
(1) For	(2) Number Filed	(3) Number Extra	(4) Rate	(5) Basic Fee \$790 (\$395)
Total Claims	25 - 20 =	5	x \$22 (x \$11)	110.00
Independent Claims	18 - 3 =	15	x \$82 (x \$41)	1,230.00
Multiple Dependent Claims			\$270 (\$135)	0.00
Assignment Recording Fee per property			\$40	40.00
			TOTAL FEE:	\$2,170.00

METHOD OF PAYMENT:

A check in the amount of the above TOTAL FEE is attached. If payment is enclosed, this amount is believed to be correct; however, the Commissioner is hereby authorized to charge any deficiency or credit any overpayment to Deposit Account No. 19-0741.

Respectfully submitted,

Date: July 22, 1998 Docket No.: 67242/107

Stephen B. Maebius

Registration No. 35,264

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

Attorney Docket No. 67242/107

In re patent application of

Fumihiko Watanabe et al.

Serial No. Unassigned

Filed: July 22, 1998

For: SULFONATED AMINO

SULFONATED AMINO ACID DERIVATIVES AND

METALLOPROTEINASE INHIBITORS CONTAINING THE SAME

PRELIMINARY AMENDMENT

Assistant Commissioner for Patents Washington, D.C. 20231

Sir:

Prior to examination of the above-identified application, Applicants respectfully request that the following amendment be entered into the application:

IN THE CLAIMS:

Claim 3, line 1, delete "or 2".

Claim 20, line 1, delete "any one of claims 4 to 19" and insert --claim 4--.

Claim 21, line 1, delete "any one of claims 4 to 7 and 10 to 19" and insert --claim 4--.

Claim 22, line 1, delete "any one of claims 4 to
19" and insert --claim 4--.

Claim 23, line 1, delete "any one of claims" and
insert --claim 4.--;

line 2, delete line in its entirety.

Attorney Docket No. 67242/107

Claim 25, line 2, delete line in its entirety
and insert --claim 4.--.

Attorney Docket No. 67242/107

REMARKS

Entry of the foregoing amendments prior to examination is respectfully requested.

Applicants respectfully request that the foregoing amendments to Claims 3, and 20-25 be entered in order to avoid this application incurring a surcharge for the presence of one or more multiple dependent claims.

Respectfully submitted,

July 22, 1998

Date

Stephen B. Maebius

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IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

ATTY. DOCKET NO. 67242/107

In re Patent Application of

Fumihiko Watanabe et al.

Serial No.: To be assigned Group Art Unit: To be assigned

Filed: July 22, 1998 Examiner: To be assigned

For: SULFONATED AMINO ACID DERIVATIVES AND METALLOPROTEINASE

INHIBITORS CONTAINING THE SAME

LETTER

Assistant Commissioner for Patents Washington, D. C. 20231

Sir:

Applicants wish to bring to the attention of the Patent Office an error in the Declaration and Power of Attorney dated July 7, 1998, of the first inventor's last name. The first inventor's last name is spelled "WANTANABE" and should be --WATANABE--, as indicated on the Assignment and the signature of Mr. Watanabe on the Declaration and Power of Attorney.

Respectfully submitted,

July 22, 1998

Date

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DESCRIPTION

SULFONATED AMINO ACID DERIVATIVES AND METALLOPROTEINASE INHIBITORS CONTAINING THE SAME

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Technical Field

This application relates to sulfonated amino acid derivatives and metalloproteinase inhibitors containing the same.

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Background Art

An extracellular matrix consists of collagen, proteoglycan, etc., has a function to support tissues, and plays a role in a maintaining of a cell functions, for example propagation, differentiation, adhesion, or the like. Matrix metalloproteinases (MMP) such as gelatinase, stromelysin, collagenase, and the like have an important role in degradation of an extracellular matrix, and these enzymes work for growth, tissue remodeling, etc. under physiological conditions. Therefore, it is considered that these enzymes participate in progression of various kind of diseases involving breakdown and fibrosis of tissues, such as osteoarthritis, rheumatoid arthritis, corneal ulceration, periodontitis, metastasis and invasion of tumor, and virus infection (for example, HIV infection). At the present time, it is not clear which enzyme participates in the above diseases seriously, but it is considered that these enzymes at least participate in tissue breakdown. As metalloproteinase inhibitors of amino acid derivatives, for example hydroxamic acid derivatives of amino acids (JP-A-6-2562939), carboxylic acid derivatives of amino acid and/or their hydroxamic acid derivatives (WO95/35276), etc. are disclosed.

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Disclosure of Invention

If it is able to inhibit the activity of MMP, it is considered that MMP inhibitors contribute to an improvement and prevention of the above diseases caused by or

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related to its activity. Therefore, development of MMP inhibitors has long been desired.

In the above situation, the inventors of the present invention found that a kind of sulfonamide derivatives have strong activity to inhibit MMP.

The present invention relates to a composition for inhibiting metalloproteinase which contains a compound of the formula \underline{I} :

$$R^5-R^4-R^3-SO_2-N$$
 COY I

wherein R^1 is optionally substituted lower alkyl, optionally substituted aryl, optionally substituted heteroaryl, or optionally substituted heteroarylalkyl; R^2 is hydrogen atom, optionally substituted lower alkyl, optionally substituted aryl, optionally substituted aryl, optionally substituted heteroaryl, or optionally substituted heteroarylalkyl; R^3 is a bond, optionally substituted arylene, or optionally substituted heteroarylene; R^4 is a bond, $\cdot(CH_2)m_-$, $\cdot CH=CH_-$, $\cdot C\equiv C_-$, $\cdot CO_-$, $\cdot CO_-$ NH \cdot , $\cdot N=N_-$, $\cdot N(R^A)_-$, $\cdot NH_-CO_-NH_-$, $\cdot NH_-CO_-$, $\cdot O_-$, $\cdot S_-$, $\cdot SO_2NH_-$, $\cdot SO_2-NH_-N=CH_-$, or tetrazol-diyl; R^5 is optionally substituted lower alkyl, optionally substituted C_3-C_8 cycloalkyl, optionally substituted aryl, optionally substituted heteroaryl, or an optionally substituted non-aromatic heterocyclic group; R^A is hydrogen atom or lower alkyl; Y is $\cdot NHOH$ or $\cdot OH$; and m is 1 or 2; provided R^2 is hydrogen atom when Y is $\cdot NHOH$, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

Mentioned in more detail, the invention relates to the following a)-b), 1)-16), and A)-C).

a) A composition for inhibiting metalloproteinase which contains a compound of the formula \underline{I} :

$$R^{5}-R^{4}-R^{3}-SO_{2}-N$$
 R^{1}
 $R^{5}-R^{4}-R^{3}-SO_{2}-N$
 R^{2}

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wherein R1 is optionally substituted lower alkyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaryl, or optionally substituted heteroarvlalkyl; R2 is hydrogen atom, optionally substituted lower alkyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaryl, or optionally substituted heteroarylalkyl; R3 is a bond, optionally substituted arylene, or optionally substituted heteroarylene; R^4 is a bond, $-(CH_2)m$ -, -CH=CH-, $-C \equiv C$ -, -CO-, -CO-NH-, -N=N-, -N(RA)-, -NH-CO-NH-, -NH-CO-, -O-, -S-, -SO₂NH-, -SO₂-NH-N=CH-, or tetrazol-diyl; R5 is optionally substituted lower alkyl, optionally substituted C3-C8 cycloalkyl, optionally substituted aryl, optionally substituted heteroaryl, or an optionally substituted non-aromatic heterocyclic group; RA is hydrogen atom or lower alkyl; Y is -NHOH or -OH; and m is 1 or 2; provided R² is hydrogen atom when Y is -NHOH, R5 is optionally substituted aryl or optionally substituted heteroaryl when R3 is optionally substituted arylene or optionally substituted heteroarylene and R^4 is -CO-NH- or -NH-CO-, R⁵ is optionally substituted anyl or optionally substituted heteroaryl when R3 is optionally substituted arylene or optionally substituted heteroarvlene and R4 is tetrazol-diyl, R5 is lower alkyl, aryl substituted by lower alkyl or optionally substituted aryl, or heteroaryl substituted by lower alkyl or optionally substituted aryl when R3 is optionally substituted arylene and R4 is a bond, both of R3 and R4 are not a bond at the same time, and R4 is not -O- when R3 is optionally substituted arylene or optionally substituted heteroarylene, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof. b) A composition for inhibiting metalloproteinase as mentioned above, which is a composition for inhibiting type-IV collagenase.

Preferred embodiment of the present invention are as follows.

1) A compound of the formula \underline{I} :

$$R^5 - R^4 - R^3 - SO_2 - N$$
 R^1
 R^2
COY I

wherein R1 is optionally substituted lower alkyl, optionally substituted aryl, optionally

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substituted aralkyl, optionally substituted heteroaryl, or optionally substituted heteroarylalkyl; R2 is hydrogen atom, optionally substituted lower alkyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaryl, or optionally substituted heteroarylalkyl; R3 is a bond, optionally substituted arylene, or optionally substituted heteroarylene; R^4 is a bond, $-(CH_2)m$ -, -CH=CH-, $-C \equiv C$ -, -CO-, -CO-NH-, -N=N-, -N(R^A)-, -NH-CO-NH-, -NH-CO-, -O-, -S-, -SO₂NH-, -SO₂-NH-N=CH-, or tetrazol-diyl; R5 is optionally substituted lower alkyl, optionally substituted C3-C8 cycloalkyl, optionally substituted aryl, optionally substituted heteroaryl, or an optionally substituted non-aromatic heterocyclic group; RA is hydrogen atom or lower alkyl; Y is -NHOH or -OH; and m is 1 or 2; provided R² is hydrogen atom when Y is -NHOH, R5 is optionally substituted aryl or optionally substituted heteroaryl when R3 is optionally substituted arylene or optionally substituted heteroarylene and ${
m R}^4$ is -CO-NH- or -NH-CO- (when R3 is phenylene and R4 is -CO-NH-, R1 is not methyl or phenyl and \mathbb{R}^5 is not 2-chlorophenyl, 4-chlorophenyl, or 2,4-dichlorophenyl), \mathbb{R}^5 is lower alkyl, optionally substituted aryl, or optionally substituted heteroaryl when R3 is optionally substituted arylene or optionally substituted heteroarylene and R4 is tetrazol-diyl, R5 is lower alkyl, aryl substituted with lower alkyl or optionally substituted aryl, or heteroaryl substituted with lower alkyl or optionally substituted aryl when R3 is optionally substituted arylene and R4 is a bond, both of R3 and R4 are not a bond at the same time, and R4 is not -O- when R3 is optionally substituted arylene or optionally substituted heteroarylene, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

2) A compound of the formula II:

$$R^{7}-R^{6} \xrightarrow{\stackrel{R^{8}}{=}} SO_{2}-N \xrightarrow{\stackrel{R^{1}}{\stackrel{}{=}}} COY \qquad \qquad \underline{II}$$

wherein R^6 is -CH=CH-, -C \equiv C-, -N=N-, -NH-CO-NH-, -S-, -SO₂NH-, or -SO₂-NH-N=CH-; R^7 is optionally substituted aryl or optionally substituted heteroaryl; R^8 and R^9

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are each independently hydrogen atom, lower alkoxy, or nitro; R¹, R², and Y are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

3) A compound of the formula III:

$$R^7 - R^{10} = SO_2 - N + COY$$

wherein R¹⁰ is -(CH₂)m-, -CO-, -CO-NH-, -N(R^A)-, -NHCO-, or tetrazol-diyl; m is 1 or 2; R¹, R², R⁷, R⁸, R⁹, R^A, and Y are as defined above, provided R¹ is not methyl or phenyl and R⁷ is not 2-chlorophenyl, 4-chlorophenyl, or 2,4-dichlorophenyl when R¹⁰ is -NH-CO-, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

4) A compound of the formula <u>IV</u>:

$$R^7 - R^{11} \longrightarrow SO_2 - N \longrightarrow COY \longrightarrow IV$$

wherein R^{11} is a bond, -CH=CH-, or -C \equiv C-; X is oxygen atom or sulfur atom, R^1 , R^2 , R^7 , and Y are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

5) A compound of the formula \underline{I} :

$$R^{5'}-R^{4'}-R^{3'}-SO_2-N$$
 COY [

wherein R¹' is benzyl, (indol-3-yl)methyl, (1-methylindol-3-yl)methyl, (5-methylindol-3-yl)methyl, (1-acetylindol-3-yl)methyl, (1-methylsulfonylindol-3-yl)methyl, (1-alkoxycarbonyl-3-yl)methyl (for example ethoxycarbonylmethyl), or i-propyl; R²' is hydrogen atom, methyl, 4-aminobutyl, or benzyl; R³' is 1,4-phenylene; R⁴' is -O-; R⁵' is phenyl or 4-hydroxy-phenyl; and Y is as defined above, its optically active substance,

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their pharmaceutically acceptable salt, or hydrate thereof.

6) A compound of the formula I":

wherein R^{1"} is 4-thiazolylmethyl, (indol-3-yl)methyl, (5-methoxyindol-3-yl)methyl, 1-naphthylmethyl, 2-naphthylmethyl, 4-biphenylylmethyl, 2,2,2-trifluoroethyl, 2-phenylethyl, benzyl, i-propyl, 4-nitrobenzyl, 4-fluorobenzyl, cyclohexylmethyl, (1-methylindol-3-yl)methyl, (5-methylindol-3-yl)methyl, (5-fluoroindol-3-yl)methyl, (pyridin-4-yl)methyl, (benzothiazol-2-yl)methyl, (phenyl)(hydroxy)methyl, phenyl, carboxymethyl, 2-carboxyethyl, hydroxymethyl, phenylmethoxymethyl, 4-carboxybenzyl, (benzimidazol-2-yl)methyl, (1-methylsulfonylindol-3-yl)methyl, or (1-ethoxycarbonylindol-3-yl)methyl; R^{2"} is hydrogen atom; R^{3"} is 1,4-phenylene; R^{4"} is a bond; R^{5"} is phenyl, 3- methoxyphenyl, 4-methoxyphenyl, 4-methylphenyl, 4-tertbutylphenyl, 4-trifluoromethylphenyl, 4-fluorophenyl, 4-methylthiophenyl, 4-biphenylyl, 2-thienyl, benzoxazol-2-yl, benzothiazol-2-yl, or tetrazol-2-yl; and Y is as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

7) A compound of the formula $\underline{\mathbf{V}}$:

$$R^7 - R^{12} \xrightarrow{R^8} SO_2 - N \xrightarrow{R^1} COOH$$
 V

wherein R^{12} is $\cdot CH = CH \cdot C = C \cdot ; R^1, R^2, R^7, R^8$, and R^9 are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof. 8) A compound of the formula \underline{VI} :

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$$R^{14} - C - N \longrightarrow R^{8} - SO_{2} - N \longrightarrow COOH$$
 VI

wherein R², R⁸, and R⁹ are as defined above, R¹³ is optionally substituted lower alkyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaryl, or optionally substituted heteroarylalkyl; and R¹⁴ is optionally substituted aryl, or optionally substituted heteroaryl; provided R¹³ is not methyl or phenyl and R¹⁴ is not 2-chlorophenyl, 4-chlorophenyl, or 2,4-dichlorophenyl, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

9) A compound of the formula VII:

$$R^{7} - N \cdot N = R^{8} \cdot R^{1} \cdot COOH$$
 VII

wherein R¹, R², R⁷, R⁸, and R⁹ are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

10) A compound of the formula VIII:

wherein R¹, R², R⁷, and R¹¹ are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

11) A compound of the formula VIII:

$$R^7-O$$
 R^8
 R^1
 R^7-O
 R^8
 R^1
 R^2
 R^2
 R^3
 R^2

wherein R1, R2, R7, R8, and R9 are as defined above, its optically active substance, their

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pharmaceutically acceptable salt, or hydrate thereof.

12) A compound of the formula \underline{X} :

$$R^7 - R^{12} - SO_2 - N - COOH X$$

wherein R^{12} is -CH=CH- or -C \equiv C-; R^1 , R^7 , R^8 , and R^9 are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

13) A compound of the formula XI:

$$R^{14}-C-N$$
 R^{10}
 R^{10}
 R^{10}
 R^{10}
 R^{10}
 R^{10}
 R^{10}
 R^{10}
 R^{10}
 R^{10}

wherein R⁸, R⁹, R¹³, and R¹⁴ are as defined above, provided R¹³ is not methyl or phenyl and R¹⁴ is not 2-chlorophenyl, 4-chlorophenyl, or 2,4-dichlorophenyl, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

14) A compound of the formula XII:

wherein R¹, R⁷, R⁸, and R⁹ are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

15 15) A compound of the formula XIII:

wherein R¹, R⁷, and R¹¹ are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

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16) A compound of the formula XIV:

$$R^7-O$$

$$= SO_2-N$$

$$= SO_2-N$$

$$= SO_2-N$$

$$= SO_2-N$$

$$= SO_2-N$$

$$= SO_2-N$$

wherein R¹, R⁷, R⁸, and R⁹ are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

A compound of the invention is more specifically illustrated below:

- A) The compound of any one of above 1) to 16), wherein R¹, R¹, R¹, and R¹³ are i-propyl, benzyl, or (indol-3-yl) methyl.
- B) The compound of any one of above 1) to 4) and 7) to 16), wherein R⁵, R⁷, and R¹⁴ are phenyl optionally substituted with one or more substituents selected from the group consisting of alkoxy, alkylthio, and alkyl.
- C) The compound of any one of above 1) to 16), wherein a configuration of asymmetric carbon atoms bonding with R¹, R¹, R¹, and R¹³ is R configuration.

Further, this invention relates to a pharmaceutical composition, a composition for inhibiting metalloproteinase, and a composition for inhibiting type IV collagenase which contain the compound above 1) to 16) and A) to C)

All of compounds of above 1) to 16) and A) to C) have strong metalloproteinase inhibitory activity, and the following compound is more preferable:

$$R^{5}-R^{4}-R^{3}-SO_{2}-N$$
 COY I

- 1) A compound wherein R^1 is i-propyl, benzyl, or (indol-3-yl) methyl, R^2 is hydrogen atom, R^3 is 1,4-phenylene, R^4 is $-C \equiv C$ -, and R^5 is optionally substituted phenyl.
- 2) A compound wherein R^1 is i-propyl, benzyl, or (indol-3-yl) methyl, R^2 is hydrogen atom, R^3 is optionally substituted 2,5-thiophen-diyl, R^4 is $-C \equiv C$ -, and R^5 is optionally substituted phenyl.
- 3) A compound wherein R1 is i-propyl, benzyl, or (indol-3-yl)methyl, R2 is hydrogen

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atom, R3 is 1,4-phenylene, R4 is tetrazol-diyl, and R5 is optionally substituted phenyl.

The term "alkyl" herein used means C₁-C₁₀ straight or branched chain alkyl, for example, methyl, ethyl, n-propyl, i-propyl, n-butyl, i-butyl, sec-butyl, tert-butyl, n-pentyl, i-pentyl, neo-pentyl, tert-pentyl, and the like.

The term "lower alkyl" herein used means C₁-C₆ straight or branched chain alkyl, for example, methyl, ethyl, n-propyl, i-propyl, n-butyl, i-butyl, sec-butyl, tertbutyl, and the like.

The term "C₃-C₈ cycloalkyl" herein used is exemplified by cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cycloheptyl, and the like.

The term "aryl" herein used means monocyclic or condensed ring aromatic hydrocarbons. Examples of the aryl are phenyl, naphthyl, and the like.

The term "aralkyl" herein used means the above mentioned alkyl substituted by the above mentioned aryl at any possible position. Examples of the aralkyl are benzyl, phenethyl, phenylpropyl (e.g., 3-phenylpropyl), naphthylmethyl (\alpha-naphthylmethyl), anthrylmethyl (9-anthrylmethyl), and the like. Benzyl is preferred. The aryl part may optionally be substituted.

The term "heteroaryl" herein used means a 5 to 6 membered aromatic heterocyclic group which contains one or more hetero atoms selected from the group consisting of nitrogen, oxygen and sulfur atoms in the ring and may be fused with a carbocyclic ring or other heterocyclic ring at any possible position. Examples of the heteroaryl are pyrrolyl (e.g., 1-pyrrolyl), indolyl (e.g., 2-indolyl), carbazolyl (e.g., 3-carbazolyl), imidazolyl (e.g., 4- imidazolyl), pyrazolyl (e.g., 1-pyrazolyl), benzimidazolyl (e.g., 2-benzimidazolyl), indazolyl (e.g., 3-indazolyl), indolizinyl (e.g., 6-indolizinyl), pyridyl (e.g., 4-pyridyl), quinolyl (e.g., 5-quinolyl), isoquinolyl (e.g., 3-isoquinolyl), acridinyl (e.g., 1-acridinyl), phenanthridinyl (e.g., 2-phenanthridinyl), pyridazinyl (e.g., 3-pyridazinyl), pyrimidinyl (e.g., 4-pyrimidinyl), pyrazinyl (e.g., 2-pyrazinyl), cinnolinyl (e.g., 3-cinnolinyl), phthalazinyl (e.g., 2-phthalazinyl), quinazolinyl (e.g., 2-quinazolinyl), isoxazolyl (e.g., 3-isoxazolyl), benzisoxazolyl (e.g., 3-benzisoxazolyl), oxazolyl (e.g., 2-oxazolyl), benzoxadiazolyl (e.g., 4-

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benzoxadiazolyl), isothiazolyl (e.g., 3-isothiazolyl), benzisothiazolyl (e.g., 2-benzisothiazolyl), thiazolyl (e.g., 2-thiazolyl), benzothiazolyl (e.g., 2-benzothiazolyl), furyl (e.g., 3-furyl), benzofuryl (e.g., 3-benzofuryl), thienyl (e.g., 2-thienyl), benzothienyl (e.g., 2-benzothienyl), tetrazolyl, and the like. The aryl part of the above heteroaryl is optionally substituted.

The term "heteroarylalkyl" herein used means the above mentioned alkyl substituted with the above mentioned heteroaryl at any possible position. Examples of the heteroarylalkyl are thiazolylmethyl (e.g., 4-thiazolylmethyl), thiazolylethyl (e.g., 5-thiazolyl-2-ethyl), indolylmethyl (e.g., 2-indolylmethyl), imidazolylmethyl (e.g., 4-imidazolylmethyl), benzothiazolylmethyl (e.g., 2-benzothiazolylmethyl), benzopyrazolylmethyl (e.g., 1-benzopyrazolylmethyl), benzotriazolylmethyl (e.g., 4-benzotriazolylmethyl), benzoquinolylmethyl (e.g., 2-benzoquinolylmethyl), benzimidazolylmethyl (e.g., 2-benzimidazolylmethyl), pyridylmethyl (e.g., 2-pyridylmethyl), and the like. The aryl part of the above heteroaryl is optionally substituted.

The term "arylene" herein used is exemplified by phenylene, naphthylene, and the like. Mentioned in more detail, it is exemplified by 1,2-phenylene, 1,3-phenylene, 1,4-phenylene, and the like.

The term "heteroarylene" herein used is exemplified by thiophen-diyl, furandiyl, pyridin-diyl, and the like, in more detail, by 2,5-thiophen-diyl, 2,5-furan-diyl, and the like.

The term "non-aromatic heterocyclic group" herein used means 5 to 6 membered non-aromatic heterocyclic group which contains one or more hetero atoms selected from the group consisting of nitrogen, oxygen and sulfur atoms in the ring, and may bind at any possible positin. Examples of the non-aromatic heterocyclic group are morpholino, piperidino, pyrrolidino, and the like.

The term "alkoxy" herein used means alkoxy of which alkyl part is the above mentioned alkyl. Examples of the alkoxy are methoxy, ethoxy, propoxy, butoxy, pentyloxy, and the like.

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The term "lower alkoxy" herein used means alkoxy of which alkyl part is the above mentioned lower alkyl. Examples of the lower alkoxy are methoxy, ethoxy, n-propoxy, i-propoxy, i-butoxy, i-butoxy, sec-butoxy, tert-butoxy, and the like.

The term "halogen" herein used means fluoro, chloro, bromo, and iodo.

The term "alkylthio" herein used means alkylthio whose alkyl part is the above mentioned lower alkyl. Examples of the alkylthio are methylthio, ethylthio, and the like.

Substituents for "optionally substituted alkyl", "optionally substituted C₃-C₈ cycloalkyl", and "optionally substituted non-aromatic heterocyclic group" are hydroxy, alkoxy (e.g., methoxy and ethoxy), mercapto, alkylthio (e.g., methylthio), cycloalkyl (e.g., cyclopropyl, cyclobutyl, cyclopentyl, and cyclohexyl), halogen (e.g., fluoro, chloro, bromo, and iodo), carboxy, alkoxycarbonyl (e.g., methoxycarbonyl and ethoxycarbonyl), nitro, cyano, haloalkyl (e.g., trifluoromethyl), substituted or unsubstituted amino (e.g., methylamino, dimethylamino, and carbamoylamino), guanidino, phenyl, benzyloxy, and the like. These substituents are able to bind them at one or more of any possible positions.

Substituents for the aromatic ring of "optionally substituted aryl", "optionally substituted aralkyl", "optionally substituted heteroaryl", "optionally substituted heteroarylalkyl", "optionally substituted arylene", and "optionally substituted heteroarylene" are, for example, hydroxy, alkoxy (e.g., methoxy and ethoxy), mercapto, alkylthio (e.g., methylthio), cycloalkyl (e.g., cyclopropyl, cyclobutyl, cyclopentyl), halogen (e.g., fluoro, chloro, bromo, and iodo), carboxy, alkoxycarbonyl (e.g., methoxycarbonyl and ethoxycarbonyl), nitro, cyano, haloalkyl (e.g., trifluoromethyl), aryloxy (e.g., phenyloxy) substituted or unsubstituted amino (e.g., methylamino, dimethylamino, diethylamino, and benzylidenamino), guanidino, alkyl (e.g., methyl, ethyl, n-propyl, i-propyl, n-butyl, i-butyl, sec-butyl, tert-butyl, n-pentyl, i-pentyl, neopentyl, and tert-pentyl), alkenyl (e.g., vinyl and propenyl), alkynyl (e.g., ethynyl and phenylethynyl), alkanoyl (e.g., formyl, acetyl, and propionyl), acyloxy (e.g., acetyloxy), acylamino, alkylsulfonyl (e.g., methylsulfonyl), phenyl, benzyl, an azo group (e.g.,

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phenylazo), optionally substituted heteroaryl (e.g., 3-pyridyl), optionally substituted ureido (e.g., ureido and phenylureido), and the like. These substituents are able to bind to it at one or more of any possible position.

Best Mode for Carrying Out the Invention

Compounds (Ia) and (Ib) of the invention are able to be synthesized from the corresponding α -amino acids represented by the formula (XV) by means of the following 6 synthetic methods. Generally, it is possible to produce the compounds of the invention by means of the method A. Each classified type of the compounds is possible to be produced by means of methods the B to F. However, these methods are only examples to produce the compounds represented by the formula I. A compound represented by the formula I produced by any other method is included in this invention.

Method A: A general synthetic method of the compound represented by the formula I.

Method B: A synthetic method of the compound wherein and R^3 is optionally substituted arylene or optionally substituted heteroarylene, R^4 is $-C \equiv C$ -, and R^5 is optionally substituted aryl or optionally substituted heteroaryl.

Method C: A synthetic method of the compound wherein R³ is optionally substituted arylene or optionally substituted heteroarylene, R⁴ is a bond, and R⁵ is optionally substituted aryl or optionally substituted heteroaryl.

Method D: A synthetic method of the compound wherein R³ is optionally substituted arylene or optionally substituted heteroarylene, R⁴ is -CO-NH-, and R⁵ is optionally substituted aryl or optionally substituted heteroaryl.

Method E: A synthetic method of the compound wherein R³ is optionally substituted arylene or optionally substituted heteroarylene, R⁴ is tetrazol-diyl, and R⁵ is optionally substituted aryl or optionally substituted heteroaryl.

Method F: A synthetic method of the compound wherein R³ is optionally substituted arylene or optionally substituted heteroarylene, R⁴ is -CH=CH-, and R⁵ is

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optionally substituted aryl or optionally substituted heteroaryl.

Details of these methods are explained as follows.

(Method A)

wherein R¹, R², R³, R⁴, and R⁵ are as defined above, R¹⁵ is hydrogen atom or a carboxy protective group, R¹⁶ is a hydroxy protective group, and Hal is halogen.

Conversion of compound (XV) to compound (Ia-1) is sulfonation of an amino group of the compound (XV) (process 1). If necessary, after this reaction, N-alkylation, deprotection of a carboxyl protective group, etc. are carried out. Conversion of compound (Ia-1) to compound (Ib-1) is to obtain hydroxamic acid derivatives from carboxylic acid derivatives (process 2). To obtain compound (Ib-1) from compound (Ia-1), compound (Ia-1) may also be reacted with hydroxylamine having a hydroxyl protective group or its acidic salts to give compound (XVI) (process 3), followed by and deprotection (process 4). Conversion to sulfonyl derivatives and hydroxamic acid derivatives are able to be carried out according to an usual method. For example, an amino acid represented by the formula (XV) is reacted with a sulfonating agent such as sulfonyl halide represented by R5-R4-R3-SO₂Hal (R3, R4, and R5 are as defined above; and Hal is halogen) and then hydroxylamine. Each process will hereinafter be described in more detail.

(Process 1)

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Some of amino acids represented by the formula (XV) or its acidic salts (e.g., hydrochloride, p-toluenesulfonate, and trifluoroacetate) which are starting materials are commercially available. The other are able to be synthesized in accordance with a method described in Zikkenkagakukoza, vol. 22, IV (nihonkagakukai), J. Med. Chem. 38, 1689-1700, 1995, Gary M. Ksander et. al., etc. some of sulfonating agents are commercially available and the other are synthesized in accordance with a method described Shin-zikkenkagakukoza, vol. 14, 1787, 1978, Synthesis 852-854, 1986, etc. A carboxyl protective group is exemplified by esters (e.g., methyl ester, tert-butyl ester and benzyl ester). Deprotection of this protective group may be carried out by hydrolysis with acid (e.g., hydrochloride and trifluoroacetic acid) or base (e.g., sodium hydroxide) depending on the type of the group, or by catalytic reduction, e.g., under 10% palladium-carbon catalyst condition. To obtain a compound (Ib-1), the esters may directly be converted to hydroxamic acid by the method of process 2. When a compound (XV) is an amino acid wherein R15 is hydrogen atom, preferable solvents for this sulfonylation are dimethylformamide, tetrahydrofuran, dioxane, dimethylsulfoxide, acetonitrile, water, or mixed solvents thereof. When a compound (XV) is an amino acid wherein R15 is a protective group such as an ester, a solvent for this sulfonylation is exemplified by the above solvents and mixed solvents of waterinsoluble solvents (e.g., benzene and dichloromethane) and the above solvents. A base to be used in this sulfonylation is exemplified by organic bases such as triethylamine, N-methylmorpholine, etc. and inorganic bases such as sodium hydroxide, potassium hydroxide, potassium carbonate, and the like. Usually this reaction can be carried out at ice-cooling to room temperature. When R1, R3, R4, R5, or R15 of compound (Ia-1) contains a functional group(s) possibly interfering this sulfonylation (e.g., hydroxy, mercapto, amino, and guanidino), it can previously be protected in accordance with a method described in "Protective Groups in Organic Synthesis" (Theodora W. Green (John Wiley & Sons)) and then deprotected at an appropriate process. When R² is not hydrogen atom, compound (Ia-1) wherein R2 is hydrogen atom is further reacted with

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haloalkyl (e.g., methyl iodide, and ethyl iodide) or haloaralkyl (e.g., benzyl chloride, and benzyl bromide) in dimethylformamide, tetrahydrofuran, dioxane, and the like at a temperature range of ice-cooling to 80 °C, preferably ice-cooling to room temperature, for 3-10 hours, preferably 10-20 hours to give the desired N-R² derivative. (Process 2)

A hydroxylamine is reacted with compound (Ia-1) or its reactive derivatives to give hydroxamic acid derivatives (Ib-1). A hydroxylamine is usually used as its acidic salts (e.g., hydrochloride, and phosphate, sulfate: commercially available) in the presence of a base. A base to be used in this reaction is exemplified by organic bases such as triethylamine, N, N-dimethylaniline, N-methylmorpholine, etc. and inorganic bases such as sodium hydroxide, potassium hydroxide, potassium carbonate, etc. When compound (Ia-1) is used as a starting material of conversion to hydroxamic acid, this reaction is carried out in the presence of a peptide condensing agent (e.g., dicyclohexylcarbodiimide, 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide, N,N'-carbonyldiimidazole, or a mixture of one of the above agents with 1-hydroxybenzotriazole, N-hydroxy sucinicimide, etc.). A solvent for this reaction may be dimethylformamide, tetrahydrofuran, dioxane, dimethylsulfoxide, acetonitrile, water, and mixed solvent thereof. This reaction is carried out at -20 °C to 40 °C, preferably ice-cooling to room temperature, for 1 to 16 hours.

Acid anhydrides (especially, mixed acid anhydrides), acid halides, acid azides, and esters can be utilized in this reaction as a reactive derivative of compound (Ia-1). These reactive derivatives are produced by usual methods. For example, the acid anhydride derivatives can be produced by a reaction of compound (Ia-1) with acid halide derivatives (e.g., ethyl chlorocarbonate) in the presence of a base (e.g., triethylamine), and acid halide derivatives can be produced by a reaction of compound (Ia-1) with a halogenation agent (e.g., oxalylchloride, and thionylchloride). Ester derivatives may be inactive or active. Sulfonyl derivatives converted from a compound (XV) wherein R¹⁵ is a carboxyl protective groups (e.g., methyl, tert-butyl, and benzyl) at process 1 can be used as inactive esters without deprotection. Active

esters can be produced by a reaction of compound (Ia-1), carbodiimide reagents (e.g., dicyclohexylcarbodiimide, 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide), and hydroxy derivatives corresponding to the active ester residue such as 1-hydroxybenzotriazole, N-hydroxysuccinimide, or the like. A reaction condition of conversion of the reactive derivatives of compound (Ia-1) to hydroxamic acid may be the same as that of conversion of compound (Ia-1) itself to hydroxamic acid. The reactions of processes 1 and 2 are able to continuously be carried out in one-pot reaction.

(Process 3)

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A protected hydroxylamine to be used in this reaction includes Obenzylhydroxylamine, O-(p-methoxybenzyl)hydroxylamine, O-(tert-butyl)hydroxylamine, or the like. This reaction condition may be in the same manner as that of process 2.

(Process 4)

This process for deprotection is carried out by catalytic reduction, treatment with conc. hydrochloric acid, or treatment with trifluoroacetic acid to give the desired compound (Ib-1). The compounds of this invention (Ia-1) and (Ib-1) can be isolated and purified by usual separation methods and purification methods (e.g., chromatography, crystallization, etc.).

20 (Method B)

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$$R^{7}-C = C-R^{17}-SO_{2}-N$$

$$R^{1}$$

$$R^{7}-C = C-R^{17}-SO_{2}-N$$

$$R^{2}$$

$$R^{7}-C = C-R^{17}-SO_{2}-N$$

$$R^{2}$$

$$R^{2}$$

$$R^{2}$$

Process 4
$$R^{7}-C = C-R^{17}-SO_{2}-N$$

$$R^{2}$$

$$\underline{Ib-2}$$
RONHOH

wherein R¹, R², R⁷, R¹⁵, and Hal are as defined above, R¹⁷ is optionally substituted aryl or optionally substituted heteroaryl.

Conversion of compound (XV) to compound (XVII) is performed by sulfonation of an amino group of compound (XV) (process 1) in the same manner as that described in process 1 of method A. Conversion of compound (XVII) to compound (XVIII) is performed by Heck reaction (K. Sonogashira, Y. Tohda, and N. Hagihara, Tetrahedron Lett., 4467(1975) etc.) wherein halogen of R¹⁷ is utilized to insert a triple bond (process 2). Conversion of compound (XVIII) to compound (Ia-2) is N-alkylation, deprotection of a carboxyl protective group, etc. (process 3), which can be carried out in the same manner as that described in process 1 of method A. Conversion of compound (Ia-2) to compound (Ib-2) is that of carboxylic acid derivatives to hydroxamic acid derivatives (process 4), which can be carried out in the same manner as those described in processes 2 to 4 of method A. Each process will hereinafter be described in more detail.

(Process 1)

This process may be carried out in the same manner as that described in process 1 of method A.

(Process 2)

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Compound (XVII) is reacted with optionally substituted aryl or optionally substituted heteroaryl having an ethynyl group such as ethynylbenzene in a solvent such as dimethylformamide, toluene, xylene, benzene, tetrahydrofuran etc. in the presence of a palladium catalyst (e.g., Pd(Ph₃P)₂Cl₂), a divalent copper reagent (e.g., CuI), and an organic base (e.g., triethylamine, and diisopropylethylamine) to give a desired compound (XVIII) (Heck reaction). This reaction is carried out at room temperature to 100 °C, preferably room temperature to 80 °C. This reaction is completed for 3 to 30 hours, preferably 10 to 20 hours. When optionally substituted aryl or optionally substituted heteroaryl has a substituent(s) interfering this reaction, the substituent(s) can previously be protected in accordance with a method of "Protective Groups in Organic Synthesis" (Theodora W. Green (John Wiley & Sons)), and then deprotected at an appropriate step.

(Process 3)

This process may be carried out in the same manner as that described in process 1 of method A.

(Process 4)

This process may be carried out in the same manner as those described in processes 2 to 4 of method A.

(Method C)

wherein R1, R2, R7, R15, R17, and Hal are as defined above.

Conversion of compound (XVII) to compound (XIX) is performed by Suzuki reaction (M. J. Sharp and V. Shieckus, Tetrahedron Lett., 26, 5997 (1985) etc.) wherein

halogen of R¹⁷ is utilized to introduce aryl or heteroaryl (process 1). Conversion of compound (XIX) to compound (Ia-3) is N-alkylation, deprotection of a carboxyl protective group, etc. (process 2) and this process can be carried out in the same manner as that described in process 1 of method A. Conversion of compound (Ia-3) to compound (Ib-3) is that of carboxylic acid derivatives to hydroxamic acid derivatives (process 3), and this process can be carried out in the same manner as those described in processes 2 to 4 of method A. Each process will hereinafter be described in more detail.

(process 1)

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Compound (XVII) is reacted with optionally substituted aryl or optionally substituted heteroaryl having a B(OH)2 (otherwise B(Et)2) group such as phenylboronic acid in a solvent such as dimethylformamide, toluene, xylene, benzene, tetrahydrofuran etc. in the presence of a palladium catalyst (e.g., Pd(Ph3P)4) and a base (e.g., potassium carbonate, calcium carbonate, triethylamine, sodium methoxide etc.) to give the desired compound (XIX) (Suzuki reaction). This reaction is carried out at room temperature to 100 °C, preferably room temperature to 80 °C. This reaction is completed for 5 to 50 hours, preferably 15 to 30 hours. When optionally substituted aryl or optionally substituted heteroaryl has a substituent(s) interfering this reaction, the substituent(s) can previously be protected in accordance with a method of "Protective Groups in Organic Synthesis" (Theodora W. Green (John Wiley & Sons)) and then deprotected at an appropriate step.

(Process 2)

This process may be carried out in the same manner as that described in process 1 of method A.

25 (Process 3)

This process may be carried out in the same manner as those described in processes 2 to 4 of method A.

(Method D)

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$$(H_{2}N_{-})R^{17} - SO_{2} - N + COOR^{15} - Process 3 - R^{7} - C - N - R^{17} - SO_{2} - N + COOR^{15}$$

$$XXI$$

$$XXII$$

$$XXII$$

Process 4

$$R^7$$
 $-C-N-R^{17}-SO_2-N$
 R^2
 $-R^2$
 $-R^3$
 R^4
Process 5

Ia-4

wherein R1, R2, R7, R15, R17, and Hal are as defined above.

Conversion of compound (XV) to compound (XX) is sulfonation of an amino group of the compound (XV) (process 1) and this process may be carried out in the same manner as that described in process 1 of method A. Conversion of compound (XX) to compound (XXI) is reduction of a nitro group of R¹⁷ to an amino group (process 2) and this process can be carried out by catalytic reduction or other reduction using hydrochloric chloride - Fe, hydrochloric chloride - Sn, etc. Conversion of compound (XXI) to compound (XXII) is performed by usual amide bond formation reaction wherein an amino group of R¹⁷ is utilized (process 3). Conversion of compound (XXII) to compound (Ia-4) is N-alkylation, deprotection of a carboxyl protective group, etc. (process 4) of compound (XXII) and this process can be carried out in the same manner as that described in process 1 of method A. Conversion of compound (Ia-4) to compound (Ib-4) is that of carboxylic acid derivatives to hydroxamic acid derivatives (process 5) and this process can be carried out in the same manner as those described

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in processes 2 to 4 of method A. Each process will hereinafter be described in more detail.

(process 1)

This process may be carried out in the same manner as that described in process 1 of method A.

(Process 2)

Compound (XX) is treated with hydrogen in a solvent such as methanol, ethanol, ethyl acetate, acetic acid, etc. in the presence of a catalyst (e.g., Pd-C, PtO₂, Raney Ni etc.), under a no-pressure or pressured condition to give the desired compound (XXI). This reaction is carried out at a temperature under ice-cooling to 80 °C, preferably room temperature to 50 °C, and is completed for 1 to 10 hours, preferably 2 to 5 hours.

(Process 3)

Compound (XXI) is reacted with optionally substituted aryl or optionally substituted heteroaryl having an acid halide (otherwise an active ester) group such as benzoyl chloride in a solvent such as dimethylformamide, tetrahydrofuran, dioxane, dimethylsulfoxide, acetonitrile, xylene, toluene, benzene, dichloromethane, etc. in the presence of a base (e.g., triethylamine, N-methylmorpholine, potassium carbonate etc.) to give the desired compound (XXII). This reaction is carried out at a temperature under ice-cooling to 100 °C, preferably room temperature to 60 °C, and is completed for 3 to 30 hours, preferably 10 to 25 hours.

(Process 4)

This process may be carried out in the same manner as that described in process 1 of method A.

25 (Process 5)

This process may be carried out in the same manner as those described in processes 2 to 4 of method A.

(Method E)

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wherein R^1 , R^2 , R^7 , R^{15} , R^{17} , and Hal are as defined above.

Conversion of compound (XV) to compound (XXIII) is performed by sulfonating an amino group of the compound (XV) (process 1) in the same manner as that described in process 1 of method A. Conversion of compound (XXIII) to compound (XXIV) is done by the reduction wherein an ethenyl group of R¹⁷ is converted into an aldehyde group (process 2). Conversion of compound (XXIV) to compound (XXVI) is performed by a tetrazole ring formation reaction (processes 3 and 4). Conversion of compound (XXVI) to compound (Ia-5) is N-alkylation, deprotection of a carboxyl protective group, etc. of compound (XXVI) (process 5), and this process can be carried out in the same manner as that described in process 1 of method A. Conversion of compound (Ia-5) to compound (Ib-5) is that of carboxylic acid derivatives to hydroxamic acid derivatives (process 6), which can be carried out in the same manner as those described in processes 2 to 4 of method A. Each process will hereinafter be described in more detail.

(process 1)

This process may be carried out in the same manner as that described in process 1 of method A.

(Process 2)

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A compound (XXIII) is treated with ozone in a solvent such as dichloromethane, ethyl acetate, methanol, etc. to form an ozonide, and then a reagent such as zinc-acetic acid, triethylphosphate, dimethylsulfide, etc. is added to this reaction mixture for reduction to give the desired aldehyde derivatives (XXIV) The reduction can also be carried out by catalytic hydrogenation. This reaction is carried out at -100 $^{\circ}$ C to room temperature, preferably -78 $^{\circ}$ C to a temperature under icecooling, and is completed for 0.5 to 10 hours, preferably 1 to 3 hours. (Process 3)

A compound (XXIV) is reacted with benzensulfonylhydrazide in a solvent such as tetrahydrofuran, ether, etc. mixed with a solvent such as methanol, ethanol, etc. to give the desired compound (XXV). This reaction is carried out at a temperature under ice-cooling to 80 $\,^{\circ}$ C, preferably room temperature to 50 $\,^{\circ}$ C, and is completed for 3 to 30 hours, preferably 10 to 20 hours.

(Process 4)

Optionally substituted aryl or optionally substituted heteroaryl having amino group such as aniline is dissolved in a mixed solvent such as alcohol (e.g., ethanol) and water. To this mixture conc. hydrochloric acid and a diazotizing agent such as a sodium nitrite aqueous solution are added at -20 °C to 10 °C, preferably 0 °C to 5 °C, to give a diazonium salt. The reaction time is 5 min to 1 hr, preferably 10 to 30 min. This reaction mixture is added to a pyridine solution of compound (XXV) and allowed react for 1 to 10 hr, preferably 2 to 5 hr, at -30 °C to 50 °C, preferably -15 °C to room temperature to give the desired compound (XXVI). When optionally substituted aryl or optionally substituted heteroaryl has a substituent(s) interfering this reaction, the substituent(s) can previously be protected in accordance with a method of "Protective Groups in Organic Synthesis" (Theodora W. Green (John Wiley & Sons)), and then deprotected at an appropriate step.

(Process 5)

This process may be carried out in the same manner as that described in

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process 1 of method A.

(Process 6)

This process may be carried out in the same manner as those described in processes 2 to 4 of method A.

5 (Method F)

$$(OHC-)R^{17}-SO_2-N + COOR^{15} + Process 1 + R^7-C=C-R^{17}-SO_2-N + COOR^{15}$$

$$\underbrace{XXIV} + XXVII$$

Process 2
$$R^7 - C = C - R^{17} - SO_2 - N$$

$$Ia-6$$

$$R^1$$
Process 3
$$R^2$$

$$R^{7}-C=C-R^{17}-SO_{2}-N$$
 R^{1}
 R^{1}
CONHOH
$$\frac{Ib-6}{R^{2}}$$

wherein R^1 , R^2 , R^7 , R^{15} , R^{17} , and Hal are as defined above.

Conversion of compound (XXIV) to compound (XXVII) is performed by Wittig reaction (G. Wittig et al., Chem. Berr. 87, 1318 (1954)) wherein an aldehyde group of R¹⁷ is utilized to introduce aryl or heteroaryl through a double bond (process 1). Conversion of compound (XXVII) to compound (Ia-6) is N-alkylation, deprotection, etc. of compound (XXVII) (process 2), and this process can be carried out the same similar as described in process 1 of method A. Conversion of compound (Ia-6) to compound (Ib-6) is that of carboxylic acid derivatives to hydroxamic acid derivatives (process 3), and this process can be carried out in the same manner as those described in processes 2 to 4 of method A. Each process will hereinafter be described in more detail. (process 1)

Compound (XXIV) is reacted with ylide derivatives of optionally substituted aryl or optionally substituted heteroaryl such as Ph₃P=CHPh, etc., which is produced

by an usual method, in a solvent such as toluene, xylene, tetrahydrofuran, ether, dimethylformamide, etc. at -100 °C to room temperature, preferably -78 °C to ice-cooling for 1 to 20 hours, preferably 1 to 5 hours, to give the desired compound (XXVII). When optionally substituted aryl or optionally substituted heteroaryl has a substituent(s) interfering this reaction, the substituent(s) can previously be protected in accordance with a method of "Protective Groups in Organic Synthesis" (Theodora W. Green (John Wiley & Sons)), and deprotected at an appropriate step.

This process may be carried out in the same manner as that described in process 1 of method A.

(Process 3)

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This process may be carried out in the same manner as those described in processes 2 to 4 of method A.

The term "compound of the present invention" herein used includes pharmaceutically acceptable salt or hydrate of the compound. The salt is exemplified by a salt with alkali metals (e.g., lithium, sodium, and potassium), alkaline earth metals (e.g., magnesium and calcium), ammonium, organic bases, amino acids, mineral acids (e.g., hydrochloric acid, hydrobromic acid, phosphoric acid, and sulfuric acid), or organic acids (e.g., acetic acid, citric acid, mallein acid, fumaric acid, benzenesulfonic acid, and p-toluenesulfonic acid). These salts can be formed by the usual method.

The compound of the present invention is not restricted to any particular isomers but includes all possible isomers and racemic modifications.

The compound of the present invention has an excellent activity for inhibiting metalloproteinase, especially activity for inhibiting MMP, and inhibits matrix dissolution, as described in the following test example. Therefore, the compound of the present invention is useful to treat or prevent diseases which are caused by MMP and relative enzymes such as TNF- α converting enzyme, etc.

Definitely, the compounds of the present invention are useful in the prevention or treatment of diseases such as osteoarthritis, rheumatoid arthritis,

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corneal ulceration, periodontal disease, metastasis and invasion of tumor, advanced virus infection (e.g., HIV), arteriosclerosis obliterans, arteriosclerotic aneurysm, atherosclerosis, restenosis, sepsis, septic shock, coronary thrombosis, aberrant angiogenesis, scleritis, multiple sclerosis, open angle glaucoma, retinopathies, proliferative retinopathy, neovascular glaucoma, pterygium, keratitis, epidermolysis bullosa, psoriasis, diabetes, nephritis, neurodegengerative disease, gingivitis, tumor growth, tumor angiogenesis, ocular tumor, angiofibroma, hemangioma, fever, hemorrhage, coagulation, cachexia, anorexia, acute infection, shock, autoimmune disease, malaria, Crohn disease, meningitis, and gastric ulcer.

When the compound of the present invention is administered to a person for treatment or prevention of the above diseases, they can be administered by oral administration such as powder, granules, tablets, capsules, pilulae, and liquid medicine, or by parenteral administration such as injections, suppository, percutaneous formulations, insufflation, or the like. An effective dose of the compound of the invention is formulated by being mixed with medicinal admixture such as excipient, penetrant, disintegrators, lubricant, and the like if necessary. When parenteral injection is prepared, the compound of the invention and an appropriate carrier are sterilized to prepare it.

An appropriate dosage varies with the conditions of the patients, an administration route, their age, their body weight and the like and should be determined by a physician in the end. In the case of oral administration, a daily dosage can generally be between 0.1 - 100 mg/kg/day, preferably 1 - 20 mg/kg/day. In the case of parenteral administration, the daily dosage can generally be between 0.01 - 10 mg/kg/day, preferably 0.1 - 1 mg/kg/day. The daily dosage can be administrated in one to several divisions.

The following examples are provided to further illustrate the present invention and are not to be constructed as limiting the scope thereof.

 $\label{eq:Abbreviations} Abbreviations \ described \ below \ are \ used \ in \ the \ following \ examples.$ $p\text{-}TsOH: p\text{-}toluene sulfonic \ acid$

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DMSO: dimethylsulfoxide

Me: methyl

^tBu: tert-butyl

Example 1 (Method A)

To a suspension of (R)-(+)-phenylalanine (compound XV-1, 1.65g (10 mmol)) in 50 ml of dimethylformamide and 35 ml of water was stirred and treated with 2.78 ml (20 mmol) of triethylamine under ice-cooling. Then, 2.52g(10 mmol) of 4-biphenylsulfonyl chloride in 10 ml of dimethylformamide was added dropwise to the mixture over 5 min. After the reaction mixture was stirred for 2 h at the same temperature, 1.35 g (10 mmol) of 1-hydroxybenzotriazole hydrate, 2.1 g (11 mmol) of 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride, 3.47 g (50 mmol) of hydroxylamine hydrochloride, and 7 ml (50 mmol) of triethylamine were added to the mixture. After being stirred for 16 h at room temperature, the reaction mixture was poured into water and extracted with ethyl acetate. The organic layer was washed with 2N HCl, 5% NaHCO₃, and water, and concentrated in vacuo. The residue was subjected to silica gel column chromatography and the fractions eluting with CHCl₃ / MeOH = 40/1 to 20/1 were collected to yield 1.70 g of compound (Ib-1-1) as a foam. Yield 43%. mp. 169-170°C.

Elemental analysis (%) C21H20N2O4S

Calcd. : C; 63.62, H; 5.08, N; 7.07, S; 8.09

Found: C;63.61, H; 5.12, N; 6.98, S; 8.06

IR ν max (cm⁻¹) (Nujol): 3365, 3295, 3266, 1674, 1320, 1159.

NMR (\delta ppm) d6-DMSO: 2.61 (dd, J=8.6, 13.4Hz, 1H), 2.80 (dd, J=6.0, 13.6Hz, 1H), 3.80

5 (m, 1H).

[α]_D: +18.5 ± 1.2 (c=0.503 %, 25°C, DMSO)

Example 1'

Another synthetic method of compound (Ib-1-1)

10 Process 1

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To a solution of (R)-phenylalanine benzyl ester tosylate (compound XV-1', 2.5 g (5.85 mmol)) in 60 ml of dichloromethane was added triethylamine (1.8 ml, 12.87 mmol) and 4-biphenylsulfonyl chloride(1.63 g, 6.44 mmol) under ice-cooling. After being stirred for 2 h at room temperature, the reaction mixture was washed with 2N HCl, 5% NaHCO₃ and water, and concentrated in vacuo. The residue was subjected to silica gel column chromatography and the fractions eluting with CHCl₃ / MeOH = 40/1 to 20/1 were collected and crystallized from dichloromethane / hexane to give 2.32 g of

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compound (Ia-1-1'). Yield 84.1%. mp. 130-131℃.

Elemental analysis (%) C28H25NO4S

Calcd. : C; 71.32, H; 5.34, N; 2.97, S; 6.80

Found: C; 71.05, H; 5.41, N; 3.00, S; 6.81

5 IR ν max (cm⁻¹) (Nujol): 3352, 1732, 1341, 1190, 1163.

NMR (δ ppm) (CDCl₃): 3.06 (d, J=5.8Hz, 2H), 4.30 (dt, J=6.0, 9.0Hz, 1H), 4.89 (s, 2H), 5.12 (d, J=9.0Hz, 1H), 6.98-7.81 (m, 14H).

 $[\alpha]_D$: -16.4 ± 1.1(c=0.506 %, 25°C, MeOH)

Process 2

A solution of compound (Ia-1-1') (2.28 g) which was obtained process 1 in 50 ml of mixed solvents of methanol / ethyl acetate =1/1, was hydrogenated using 10 % Pd/C (200 mg) for 25 min. The reaction mixture was filtered off, and the filtrate was concentrated in vacuo. The residue was recrystallized from dichloromethane / hexane to give 1.83 g of compound (Ia-1-1"). Yield 99.1 %. mp. 146-147°C.

Elemental analysis (%) C21H19NO4S

Calcd.: C; 66.12, H; 5.02, N; 3.67, S; 8.41

Found: C;65.97, H; 5.06, N; 3.61, S; 8.48

IR ν max (cm⁻¹) (Nujol): 3408, 3305, 1751, 1325, 1161, 1134.

NMR (δ ppm) (CDCl₃): 2.97 (dd, J=7.0, 13.8Hz, 1H), 3.14 (dd, J=5.2, 14.0Hz,1H), 4.13 (m, 1H), 7.03-7.78 (m, 14H).

 $[\alpha]_{D}$: -4.0±0.4(c=1.000 %, 25°C, MeOH)

Process 3

To a solution of compound (Ia-1-1", 1.0 g (2.62 mmol)) which was obtained process 2 in dichloromethane (20 ml) was added 0.33 ml (3.93 mmol) of oxalyl chloride and one drop of dimethylformamide. After being stirred for stirred for 1 h at room temperature, the reaction mixture was concentrated in vacuo. The residue was dissolved in 10 ml of tetrahydrofuran. A solution of hydroxylamine hydrochloride (911 mg (13.1 mmol)) and NaHCO₃ 1.54 g (18.34 mmol) in 10ml of tetrahydrofuran and 10ml of water was stirred for 5 min under ice-cooling. To the mixture was added the

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above solution of acid chloride in tetrahydrofuran and the resulting mixture was stirred for 30 min. The reaction mixture was poured into water, and extracted with ethyl acetate. The organic layer was washed with 5% NaHCO₃, and water, and concentrated in vacuo to give compound (Ia-1) (969 mg). Yield 93.3 %.

5 Process 4

To a solution of compound (Ia-1-1", 2.0 g, 5.24 mmol) which was obtained process 2 in dimethylformamide (20 ml) was added 1-hydroxybenzotriazole hydrate (0.7 g, 5.24 mmol), N-methylmorpholine (2.9 ml, 26.2 mmol), 1-ethyl-3-(3-diisopropylamino) carbodiimide hydrochloride (8 mmol), and O-benzylhydroxylamine hydrochloride (1.67 g, 10.48 mmol), and the resulting mixture was stirred for 6 h at room temperature. The reaction mixture was poured into water and extracted with ethyl acetate. The organic layer was washed with 2N HCl, 5% NaHCO₃, and water, and concentrated in vacuo. The residue was subjected to silica gel column chromatography and the fractions eluting with CH₂Cl₂ / hexane = 1/1 were collected and recrystallized from dichloromethane / hexane to give 2.04 g of compound (XVI-1). Yield 80 %. mp. 171-173°C.

Elemental analysis (%) C28H26N2O4S

Calcd.: C; 69.12, H; 5.39, N; 5.76, S; 6.59

Found: C; 68.85, H; 5.46, N; 5.76, S; 6.78

20 IR ν max (cm⁻¹) (Nujol): 3248, 1661, 1594, 1333, 1163.

NMR (δ ppm) (CDCl₃): 2.85-3.60 (m, 2H), 3.86 (m, 1H), 4.77 (ABq-Apart, J=11.4Hz, 1H), 4.82 (ABq-Bpart, J=11.4Hz, 1H), 5.00 (m, 1H), 6.95-7.70 (m, 19H).

 $[\alpha]_D$: -40.2 ± 1.6 (c=0.505 %, 25°C, DMSO)

Process 5

A solution of compound (XVI-1) (1.97 g) which was obtained process 4 in a 60 ml of mixed solvents of methanol / ethyl acetate =1/1 was hydrogenated using 10 % Pd-C (200 mg) for 3.5 h. The reaction mixture was filtered off, and the filtrate was concentrated in vacuo. The residue was recrystallized from dichloromethane / hexane to give 1.35 g of compound (Ib-1-1). Yield 84.4-%.

Example 2 - 91

The compounds which were shown in Tables 1 to 22 were synthesized in a manner similar to those described in Example 1'

			:		(ai)	
Example No.	<u>~</u>	R -8	*	mp (decomp.)	IR (v cm·1) (KBr)	1H-NMR(& ppm) ds-DMSO
2	S N ———CH ₂ -		RS	173 >	3258,1650,1377, 1348,1163 (Nujol)	2.87(dd,J=5.6,14.2Hz,1H), 2.98(dd, J=8.4,14.2Hz,1H),4.02(dd,J=2.2, 8.6Hz,1H), 7.24(d,J=2.0Hz,1H), 8.83(d,J=2.2Hz,1H)
3	CH ₂ .		R	203-206	3403,3386,3265,1673 ,1320,1162 (Nujol)	2.72(dd,J=7.2,13.8Hz,1H), 2.97(dd, 7.0,14.8Hz,1H),3.81(m,1H),
4	H ₃ CO		RS	1	l	1
5	Ş.		RS	124-126	3277,1669,1397, 1322,1159,	3.12(dd,J=10.3,14.3Hz,1H), 3.89(dd, J=3.3,13.5Hz,1H),4.20(m,1H), 5.90 (brs,1H)
9	CH2-CH2-		R	139-141	3262,1663,1322, 1157,	2.67(dd,J=9.2,13.1Hz,1H), 2.84(dd, J=5.3,13.5Hz,1H),3.82(m,1H)
2	CF ₃ CH ₂ -		R	167-169	3265,1676,1642, 1337,1161 (Nujol)	2.2-2.7(m,2H),3.99(t,J=7.0Hz,1H)
∞	-ZH2CH2-		SS.	172-173	3403,3261,1669, 1321,1160	1.68(m,2H), 2.37(m,2H), 3.64(t, J=6.9Hz,1H)
6	-cH2-	Br	24	144-146	3700-2200br,3264, 1635,1342,1164,	2.61(dd,J=9.4,13.8Hz,1H),2.78(dd, J=6.0,13.8Hz,1H),3.78(m,1H),7.43 (d,J=8.2Hz,2H),7.60(d,J=8.2Hz,2H),

	¹ H-NMR(δ ppm) de-DMSO	2.60-2.82(m,2H),3.84(m,1H),7.00-7.18(m,5H),7.62-7.80(m,4H),	2.70-2.93(m,2H),2.82(s,6H), 3.75(m,1H),	0.71(d,J=6.8Hz,3H),0.74(d,J=5.4Hz,3H),1. 73(m,1H),1.73(m,1H),3.22(m,1H),3.82(s,3 H),7.05(d,J=9.0Hz,2H),7.69(d,J=9.0Hz,2H)	2.80(dd,J=10.0,13.8Hz,1H),2.92(dd, J=5.0,12.8Hz,1H),3.90(dd,J=5.4, 9.6Hz,1H),	2.62(dd,J=9.9,13.5Hz,1H),2.78(dd, J=5.8,13.0Hz,1H),3.77(t,J=6.2Hz, 1H),	0.50-1.62(m,13H), 3.56(t,J= 7.4Hz,1H)	2.71(dd,J=7.9,14.2Hz,1H),2.94(dd, J=6.9,14.2Hz,1H),3.57(s,3H),3.83 (dd,J=7.0,7.4Hz,1H)	2.25(s,3H),2.67(dd,J=7.5,14.2Hz, 1H),2.95(dd,J=7.7,14.6Hz,1H), 3.81(dd,J=6.2,14.2Hz,1H)
(a) HOH	IR (v cm·l) (KBr)	3600-2400br,3257, 1743,1721,1323,1132,	3700-2100br,3176, 1664,1320,1143,	3268,1632,1598, 1336,1162	3257,1662,1516, 1344,1322,1160,	3258,1669,1509, 1322,1157	3278,2920,1632, 1337,1161	3272,1631,1332, 1161	3404,1670,1320, 1159
R¹ R¹®SO₂NH *CONHOH	mp (decomp.) (C)	116-118	91-92	178-179	184-185	128-130	165-166	172-173	144-146
R ¹⁸ .S	*	æ	R	æ	RS	SS	×	RS	RS
	R 18	F ₃ C	(CH ₃) ₂ N	H3CO					
	<u>~</u>	CH2-CH2-	CH2-CH2-	(CH ₃) ₂ CH—	O ₂ N CH ₂ -	F CH2-CH2-	CH2-	CH3 CH3	H ₃ C
	Example No.	10	1 1	1.2	1 3	14	1.5	1 6	1 7

	1H.NMR(& ppm) ds-DMSO	2.72(dd,J=8.0,14.0Hz,1H),2.90(dd, J=6.2,14.2Hz,1H),3.82(m,1H)	1	2.68(dd,J=9.8,13.7Hz,1H),2.79(dd, J=5.6,12.8Hz,1H),3.85(t,J=7.0Hz,1H),	3.22-3.38(m,2H),4.17-4.24(m,2H), 7.80(d,J=8.0Hz,2H),7.96(d,J=6.4 Hz,2H)	3.86(d,J=3.6Hz,1H),4.91 (d,J=3.6Hz,1H)	4.88(d,J=9.4Hz,1H),8.74(d,J=9.4Hz,1H), 8.98(s,1H),10.92(s,1H)	2.69(dd,J=7.6,13.5Hz,1H),2.93(dd, J=7.6,13.5Hz,1H),3.77(t,J=7.6Hz, 1H),(CD ₃ OD)	2.74(dd,J=8.3,13.5Hz,1H),2.95(dd, J=6.5,13.5Hz,1H),3.87(dd,J=6.5, 8.3Hz,1H),(CD ₃ OD)
(2.1)	IR (v cm·¹) (KBr)	3420,1670,1592, 1321,1159	1	3186,1593,1480, 1379	3700-2400br,3252, 1668,1326,1160	3455,3362,1672, 1398,1162	3404,3315,1669, 1594,1316,1162	3700-2400(br),3473, 1675,1310,1152	3700-2200(br),3278, 1706,1645,1322,1162
	mp (decomp.)	I	1	154-158	111-115	1	196-197	197-199	201-202
	*	RS	RS	RS	RS	RS	R	24	24
	R 18							НО	ноос-
	R. 1	L CH2.	CH ₂ .	N CH2-	S CH2-			CH2-	-ZHO-CH ₂ -
	Example No.	1 8	1 9	2.0	2.1	2.2	2.3	2.4	2.5

					(ai)	
Example No.	R.	R 18	*	mp (decomp.) (C)	IR (v cm·1) (KBr)	1H-NMR(ô ppm) d ₆ -DMSO
26	CH2-		æ	63-65	3700-2200(br),3362,1670, 1590,1336,1152	2.60(dd,J=9.0,13.8Hz,1H),2.79(dd, J=9.3,13.8Hz,IH),3.76(m,1H)
2.7	CH ₂ .	O ₂ N-{}	В	70-71	3700-2200br,3372,1674, 1531,1348,1310,1161	2.66(dd,J=9.5,13.6Hz,1H),2.79(dd,J=5. 4,13.6Hz,1H),3.84(m,1H),7.73(A ₂ BzqJ= 8.9Hz,2H),8.20(A ₂ Bzq,J=8.9Hz,2H),8.7 2(d,J=9.0Hz,1H),8.86(s,1H),10.7(s,1H)
2.8	HOOC-CH ₂ -		24	ļ	l	-
2.9	HOOC-CH ₂ -CH ₂ -		R	l	1	-
3.0	HOCH ₂ -		24	192-193	3700-2400(br),3392, 1667,1320,1161	3.29(dd,J=5.7,10.7Hz,1H),3.43(dd,J= 8.4,10.7Hz,1H),3.62(m,1H),7.85(A ₂ B 2q,J=8.7Hz,2H),7.88(A ₂ B ₂ q,J=8.7Hz, 2H),7.98(d,J=7.8Hz,1H),10.61(s,1H)
3.1	CH2OCH2-		24	02-69	3700-2200(br),1671, 1329,1163	2.69(dd,J=7.6,13.5Hz,1H),2.93(dd, J=7.6,13.5Hz,1H),3.77(t,J=7.6Hz, 1H),(CD ₃ OD)
3.2	ноос-{}-сн ₂ -		×	l	l	l
3 3	CH ₂ .		~	160-162	3401,3260,1673, 1316,1165	2.66(dd,J=7.5,13.4Hz,1H),2.96(dd, J=7.6,14.2Hz,1H),3.81(m,1H)

Table 5

1			
	¹ H-NMR(δ ppm) ds-DMSO	I	2.84-3.21(m,2H),4.29(m,1H)
(а) нон	IR (v cm·l) (KBr)	1	3700-2400(br),1672, 1443,1327,1094
R¹8SO ₂ NH *CONHOH	mp (decomp.)	I	141-145
H ¹⁸ -S	*	R	RS
_	R -8	Br (S)	
	R I	TZ B	ZI ZI ZI
	Example No.	3.4	3.5

	(13)
<u>.</u>	 DIB. CO. NIL

	·					-		
¹ H-NMR(δ ppm) ds-DMSO	2.95(dd,J=9.0,14.0Hz,1H),3.12(dd, J=5.4,14.0Hz,1H),4.13(m,1H),7.29 (d,J=2.0Hz,1H),8.34(d,J=8.6Hz,1H),8.88(d,J=2.0Hz,1H),12.79(br,1H)	2.88(dd,J=8.0,14.0Hz,1H),3.09(dd, J=6.0,14.0Hz,1H),3.91(m,1H),8.23 (m,1H),10.79(s,1H),12.70(br,1H)	2.75-3.06(m,2H),3.69(s,3H),3.90 (m,1H)	3.17(dd,J=7.4,13.8Hz,1H),3.57(dd, J=5.5,13.9Hz,1H),3.80(t,J=5.6Hz, 1H),8.11(d,J=7.4Hz,1H)	2.77(dd,J=9.7,13.7Hz,1H),3.03(dd, J=4.9,13.3Hz,1H),3.93(m,1H),8.38 (d,J=8.8Hz,1H)	2.40-2.90(m,2H),4.05(m,1H),8.51 (d,J=9.0Hz,1H),13.2(br,1H)	1.83(m,2H),2.52(m,2H),3.70(m, 1H),8.32(d,J=9.0Hz,1H)	2.86(m,1H),2.87(s,6H),2.98(dd,J= 5.1,13.8Hz,1H),4.15(m,1H),5.54 (m,1H)
IR (v cm ⁻¹) (KBr)	3276,2503br,1897br, 1724,1344,1170(Nujol)	3386,3305,1747,1363, 1323,1161,1135(Nujol)	2400-3700(br),1734, 1484,1327,1160	3446,3065,1594,1397, 1303,1154,1094	3184,1723,1337, 1317,1156	3276,1706,1344, 1260,1165	3289,1739,1326, 1159,1089	2200-3700br,3439,3288, 1725,1329,1143
mp (decomp.) (C)	159-161	227-229	181-189	198-200	213-215	176-177	153-156	103-105
*	RS	ж	RS	RS	R	R	RS	R
R 18								N ² (CH ₂)
R.¹	S N CH2-	H CH ₂ -	H ₃ CO H	₩	CH2-CH2-	CF ₃ CH ₂ -	CH2CH2-	CH2-
Example No.	2	3	4	5	9	7	8	1.1

	¹ H-NMR(δ ppm) d ₆ -DMSO	2.86(dd,J=10.2,13.2Hz,1H), 3.14(dd,J=4.5,13.7Hz,1H), 4.02(m,1H),8.42(d,J=8.4Hz,1H)	2.71(dd,J=9.9,13.7Hz,1H),2.96(dd, J=5.3,13.5Hz,1H),3.89(m,1H), 8.34(d,J=9.0Hz,1H)	0.52-1.72(m,13H),3.68(m,1H), 8.20(br.s,1H)	2.80-3.12(m,2H),3.61(s,3H), 3.94(m,1H),8.30(d,J=8.6Hz,1H)	2.28(s,3H),2.78-3.10(m,2H),3.91 (m,1H),8.29(d,J=8.3Hz,1H)	2.80-3.10(m,2H),3.92(m,1H), 8.29(d,J=8.2Hz,1H)	2.60-3.04(m,2H),3.98(m,1H)	3.24-3.56(m,2H),4,34(m,1H)
ООН (Іа)	IR (v cm ⁻¹) (KBr)	3113,1724,1520, 1345,1158	3426,3114,1715, 1509,1224,1159	2919,1688,1448, 1335,1326,1169	3432,3294.1713, 1482,1341,1159	3419,3397,3291,1736, 1482,1336,1321,1165	3407,3285,1751,1735, 1703,1486,1321,1162	2600-3700br,1635,1594, 1335,1163,1095	2200-3700br,1713br, 1345,1125
R¹ R¹®·SO₂NH *_COOH	mp (decomp.) (C)	212-213	164-165	85-87	179-183	115-120	208-211	197-205	196-199
ar E	*	RS	RS	æ	RS	RS	RS	RS	RS
	R 18								
	. R	O ₂ N CH ₂ -	F CH2-	CH2-	CH ₂ -CH ₂ -	H ₃ C	F CH ₂ .	N_CH2-	S CH2-
	Example No.	1 3	1 4	1.5	1 6	1.7	1 8	2 0	2 1

R¹ B¹⁸⁻SO₂NH + COOH (la)

Example No.	٦. ۱.	R -8	*	mp (decomp.) (C)	IR (v cm ⁻¹) (KBr)	¹ H-NMR(δ ppm) d _c -DMSO
2.2	10-Ö		RS	141-143	3335,3246,1732, 1315,1152	4.10(d.J=3.2Hz,1H),5.13(d,J= 3.2Hz,1H)
23			24	211-214	3316,1734,1325, 1159(Nujol)	4.94(d,J=9.4Hz,1H),8.80(d,J= 9.4Hz,1H),13.0(br.s,1H)
2.8	H00C-CH ₂ -		æ	171-173	3353,1752,1326, 1155,1096	2.45(dd,J=6.2,16.4Hz,1H)2.63(dd, J=6.6,16.4Hz,1H),
2.9	HOOC-CH ₂ -CH ₂ -		24	185-187	3270.1709,1336, 1159,1093	1.68(dd,J=7.9,14.1Hz,1H),1.87(dd, J=6.0,13.4Hz,1H),2.22(t,J=7.2Hz, 2H),3.80(m,1H),
3.0	HOCH ₂ -		22	277-279	2200-3700br,3430, 3292,1728,1324,1162	3.51(dd,J=6.0,12.9Hz,1H),3.55(dd, J=5.4,12.9Hz,1H),3.80(m,1H), 8.06(d,J=8.7Hz,1H)
3.1	CH ₂ OCH ₂ .		æ	89-91	2200-3700br,3432, 3289,1733,1330,1165	3.54(dd,J=4.8,9.9Hz,1H),3.60(dd, J=5.7,9.9Hz,1H),4.04(m,1H), 4.39(s,2H),8.34(d,J=8.1Hz,1H)
3.2	H00C		æ	>270	3319,3052,1701,1317, 1284,1162	2.81(dd,J=9.7,13.7Hz,1H),3.05(dd, J=4.8,13.4Hz,1H),3.96(m,1H), 8.40(d,J=9.0Hz,1H),12.88(br.s,1H)

Table 9

			
	'H-NMR(δ ppm) de-DMSO	3.06(dd,J=5.4,14.4Hz,1H),3.14(dd, J=5.1,14.4Hz,1H),3.65(t,J=5.4Hz, 1H),6.92(m,1H),10.72(s,1H)	3.17-3.50(m,2H),4.51(m,1H)
ООН (Іа)	IR (v cm·1) (KBr)	3420,1588,1402, 1324,1151	2200-3700br,1734, 1334,1161
R¹ L¹8·SO ₂ NH ★COOH (Ia)	mp (decomp.)	243-246	151-156
H.	*	22	RS
	R - 8	Br As	
	R.¹	IZ TO	Z= NI NI NI
	Example No.	3.4	3 5

-									
	Elemental analysis	1	-	C ₂₄ H ₂₂ N ₂ O ₅ S+0.5H ₂ O Calc. C:62.73 H:5.04 N:6.10 S:6.98 Foun.C:62.75 H:5.08 N:6.31 S:7.05	C ₂₄ H ₂₂ N ₂ O ₅ S*0.8H ₂ O Calc. C:62.00 H:5.12 N:6.03 S:6.90 Foun.C:62.03 H:5.06 N:6.08 S:6.82	I	-	l	C ₁₇ H ₁₉ NO ₄ S-0.1CF ₃ COOH Calc. C:59.99 H:5.58 N:4.06 S:9.30 Foun.C:60.37 H:5.74 N:4.13 S:9.76
	IR (\(\nu\) cm ⁻¹) (KBr)	1726,1354 1326,1161	1732,1594 1404,1155	1607,1594 1294,1153	1724,1594 1326,1159	1685,1349 1166	1725,1599 1372,1173	1745,1653 1391,1147	1714,1594 1334.1166
	mp (decomp.) (C)	>145	_	188-190	90-93	149-152	104-107	167-169	155-157
,	*	RS	RS	R	R	R	R	24	22
	R18			H3CO-{}	H3CO	H3C		H ₃ CS	
	R 1	SO ₂ CH ₃	COOC2H ₅	LY CH	IN OH	CH ₂ .	IN OH2-	OH ₂ .	(CH ₃) ₂ CH-
	Example No.	36	3.7	3 8	3.9	4 0	4 1	4 2	4 3

Table 11

			· · · · ·						
	Elemental analysis	C ₂₁ H ₂₇ NO ₄ S•0.3H ₂ O Catc. C:63.87 H:7.04 N:3.55 S:8.12 Foun.C:63.84 H:6.86 N:3.42 S:8.01	C ₂₃ H ₂₃ NO ₄ S-0.3H ₂ O Calc. C.66.58 H:5.73 N:3.38 S:7.73 Foun.C.66.45 H:5.52 N:3.24 S:7.56	1	Ī	C ₁₇ H ₁₈ FNO ₄ S Calc. C:58.11 H:5.16 F:5.41 N:3.99 S:9.12 Foun.C:58.11 H:5.17 F:5.86 N:3.92 S:9.69	1	-	C ₂₇ H ₂₃ NO ₄ S _* 0.7H ₂ O Calc. C:68.98 H:5.23 N:2.98 S:6.82 Foun.C:69.08 H:5.09 N:2.91 S:6.73
(t	IR (v cm·1) (KBr)	1724,1340 1328,1167	1734,1719 1324,1160	1670,1375 1148	1717,1694 1349,1165	1634,1334 1158	1681,1319 1162	1725,1340 1159	1750,1324 1159
R ¹⁸ ·SO ₂ NH´ * COOH (Ia)	mp (decomp.)	196-197	241-243	157-159	175-176	145-147	183-186	183-184	224-226
NZ NH	*	R	R	В	R	æ	æ	æ	æ
R ¹⁸ -S(R 18	¹Bu-{}		F ₃ C	H3CO	F-{}-{}-	H3C	H3000H	
	- X	(CH ₃) ₂ CH-	(СН ₃)2СН-	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-	CH2-	CH2-
	Example No.	4 4	4.5	4 6	4.7	4 8	4 9	5.0	5 1

Table 12

		. 1		τ		- 1		
Elemental analysis	1	-	į	C ₁₈ H ₂₁ NO ₄ S _{2*} 0.2H ₂ O Calc. C:56.43 H:5.63 N:3.66 S:16.74 Foun.C:56.74 H:5.67 N:3.86 S:16.35		-	C ₂₁ H ₁₈ N ₂ O ₄ S _{2*} 0.3H ₂ O Calc. C:58.40 H:4.34 N:6.45 S:14.85 Foun.C:58.40 H:4.44 N:6.58 S:14.57	C ₁₇ H ₁₄ ClN ₃ O ₆ S-0.3H ₂ O Calc. C:47.48 H:3.44 Cl:8.39 N:9.65 S:7.52 Foun.C:47.57 H:3.43 Cl:8.26 N:9.79 S:7.47
IR (v cm·¹) (KBr)	1685,1349 1166	1691,1567 1390,1159	1749,1592 1323,1164	1746,1337 1164	1649,1337 1165	1588,1308 1141	1744,1592 1323,1160	1751,1734 1537,1347 1172
mp (decomp.)	157-160	111-112	194-195	197-199	108-110	187-190	239-243	222-224
*	æ	R	ж	R	R	R	R	æ
ж 8-	H3C		H3CS-{}	H ₃ CS	-{}ооон	(H ₃ C) ₂ N (H ₃ C)		N ² O
. X	CH ₂ -	CH2-CH2-	CH ₂ -	(CH ₃) ₂ CH-	CH ₂ .	CH ₂ .	COOC ₂ H ₅	CH ₂ -
Example No.	5.2	5 3	5.4	5 5	5 6	5.7	2 8	5 9
	R 1	R^{-1} R^{-18} * mp (decomp.) (KBr) (C) (KBr) A_3C R R R R R	R^{1} R^{18} * mp (decomp.) (KBr) (C) (KBr) (C) (KBr) (C) (KBr) (C) (KBr) (C) <t< td=""><td>R^{1} R^{18} * mp (decomp.) (KBr) C <</td><td>R^{1} R^{18} * $mp (decomp.)$ $IR (v cm^{-1})$ C C</td><td>$\begin{array}{cccccccccccccccccccccccccccccccccccc$</td><td>$\begin{array}{cccccccccccccccccccccccccccccccccccc$</td><td> R R R C R R R R R R</td></t<>	R^{1} R^{18} * mp (decomp.) (KBr) C <	R^{1} R^{18} * $mp (decomp.)$ $IR (v cm^{-1})$ C	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	R R R C R R R R R R

) H-NMR(& ppm) d ₆ -DMSO	2.60(dd,J=8.7,13.7Hz,1H), 2.79(dd, J=6.0,13.7Hz,1H),3.75(ddd,J=6.0, 8.7,9.0,1H),6.94(d,J=8.9Hz.2H)	2.71(dd,J=7.0,14.4Hz,1H), 2.96(dd, J=7.0,14.2Hz,1H),3.78(t,J=7.6Hz, 1H)	2.71(dd,J=7.9,14.4Hz,1H),2.96(dd, J=7.6,14.4Hz,1H),3.78(dd,J=7.2, 7.3Hz,1H)	584, 0.76(d,J=6.6Hz,6H), 1.77(m,1H), 3.26(m,1H)	582, 2.71(dd,J=7.9,14.2Hz,1H),2.93(dd, J=6.5,14.3Hz,1H),3.65(s,3H),3.78 (dd,J=7.1,7.2Hz,1H)	2.34(s,3H),2.65(dd,J=7.8,14.1Hz, 1H),2.93(dd,J=7.6,14.4Hz,1H), 3.75(dd,J=6.8,7.7Hz,1H)	2.71(dd,J=8.9,14.4Hz,1H),2.89(dd, J=6.6,14.4Hz,1H),3.75(dd,J=6.5, 6.8Hz,1H)	6,1582, 2.54(s,3H),2.69-2.89(m,2H),3.87 (m,1H)
(qi) HOH	IR (v cm ⁻¹) (KBr)	3700-2400br,3277, 1669,1325,1152	3302,1667,1324, 1153(Nujol)	3406,1670,1582, 1325,1153	3268,1634,1584, 1336,1157	3314,1669,1582, 1420,1328,1154	3405,1671,1582, 1487,1324,1154	3317,1670,1582, 1488,1323,1153	3421,1702,1676,1582, 1354,1328,1153
R ¹⁸ SO ₂ NH CONHOH	mp (decomp.)	foam	115-118		149-151	١		1	l
R ¹⁸ .5	*	R	æ	Ω	22	SS.	RS	RS	RS
	R -8	000					000	000	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\
	R 1	CH2-	TX ST	Ę,	(CH ₃) ₂ CH-	CH3.	N N CH	1 ()	COCH ₃
	Example No.	0 9	6 1	6.2	6 3	6.4	6 5	99	2.9

Table 14

)Н (la)	'H-NMR([§] ppm) d ₆ -DMSO	2.72(dd,J=8.7,13.6Hz,1H),2.94(dd, J=5.6,13.6Hz,1H),3.84(ddd,J=5.6, 8.7,8.7Hz,1H),8.23(d,J=8.7Hz,1H)	2.88(dd,J=7.4,15.2Hz,1H),3.07(dd, J=6.2,14.4Hz,1H),3.83(m,1H),8.08 (m,1H),10.80(s,1H),12.70(br,1H)	2.81-3.12(m,2H),3.88(m,1H),8.19 (d,J=8.4Hz,1H)	0.89(d,J=7.0Hz,3H),0.98(d,J=6.8 Hz,3H),2.12(m,2H),3.80(dd,J=4.7 ,9.7Hz,1H),5.17(d,J=9.6Hz,1H)	2.78-3.10(m,2H),3.67(s,3H), 3.86(m,1H)	2.34(s,3H),2.75-3.08(m,2H),3.86(m,1H), 8.19(d,J=8.4Hz,1H)	2.78-3.08(m,2H),3.85(m,1H),8.18 (d,J=8.6Hz,1H)	2.55(s,3H),2.79-3.11(m,2H),3.98 (m,1H)
	IR (\(\nu\) cm ⁻¹) (KBr)	2400-3600br,3345,3213, 1735,1700,1346,1163	3410,3276,1724,1582, 1488,1331,1152(Nujol)	3412,1724,1582,1488, 1332,1152	3154,1720,1688,1583, 1488,1251	3273,1724,1582,1487. 1331,1198,1153	3409,3281,1725,1582, 1331,1197,1153	3415,1725,1582,1488, 1329,1196,1174,1152	3296,1742,1647,1604, 1581,1342,1334,1152
R¹ R¹®·SO₂NH <mark>↑</mark> COOH	mp (decomp.)	108-109	82-87	foam	137-138	1	1	l	236-237
R ¹⁸	*	R	R	w	x	RS	RS	SS.	RS
	R 18	0.0	₽	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	₽			\bigcirc	⟨ }∘⟨⟩
	- N	CH2-CH2-	ZZ CF2-	TN TO	(CH ₃) ₂ CH-	O-St.	H ₃ C	TZ HÖ	COCH ₃
	Example No.	0.9	6.1	6.2	6 3	6.4	6 5	9 9	29

Table 15

R ¹⁸ -SO ₂ NH ★ COOH (la)	Elemental analysis	l	C ₂₄ H ₂₂ N ₂ O ₇ S ₂ Calc. C:56.02 H:4.31 N:5.44 S:12.46 Foun.C:55.75 H:4.40 N:5.41 S:12.21	-
	mp (decomp.) IR (\(\nu\) cm ⁻¹) (KBr)	1608,1590 1507,1232 1157	1735,1583 1362,1171	1733,1583 1150
	mp (decomp.) (C)	>240	I	l _
N ₂ NH	*	В	RS	RS
R ¹⁸ -SO ₂ NI	 	-{}-0-{}-ОН	√ >• √)	√ >• √)
	. A	TZ H	SO ₂ CH ₃	COOC2Hs N N CH2-CH2-
	Example No.	8 9	6 9	0.2

	¹ H-NMR([§] ppm) d ₆ -DMSO	0.90(t,J=6.8Hz,3H),1.22-1.40(m,4H),1.52-1.6 7(m,2H),2.62(t,J=7.7Hz,2H),2.86(dd,J=8.4,13 .7Hz,1H),3.02(dd,J=5.7,13.7Hz,1H) (CDCl ₃)	0.87(t,J=6.3Hz,3H),2.50(t,J=7.4Hz,2H) 2.76(dd,J=9.6,14.0Hz,1H),2.87(dd,J=5 8,14.0Hz,1H),3.84(dd,J=5.8,9.6Hz,1H)	0.79(t,J=7.0Hz,3H),2.32-2.56(m,2H), 2.92(m,1H),3.26(m,1H),	-	2.80(m,1H),2.96(m,1H),3.94(s,2H),3.86(m,1H),6.80-7.52(m,10H),7.08(A ₂ B ₂ qJ=7 5Hz,2H),7.42(A ₂ B ₂ q,J=7.5Hz,2H)(CDC ₁₃	1
(al) HOH	IR (\(\nu\) cm ⁻¹) (KBr)	3700-2400br,3247, 1636,1337,1160	3700-2400br,1663, 1320,1145 (film)	3600-2400br,3262,1673, 1321,1142 (CHCl ₃)	_	3700-2200(br),3262, 1639,1332,1156	I
R¹ R¹®SO₂NH *CONHOH	mp (decomp.)	129-131	lio	oil	l	85-86	I
H ¹⁸ .9	*	24	R	R	R	R	R
	R 18	CH ₃ (CH ₂)4	CH ₃ (CH ₂) ₇ —	CH ₃ (CH ₂) ₃ —	CI CH3		
		-ZH2-	CH2-	CH2-	IZ HO	CH2-	LX N
	Example No.	7.1	7.2	7.3	7.4	7.5	9 2

Table 17

¹ H-NMR(δ ppm) ds-DMSO	2.79(dd,J=8.5,13.4Hz,1H),2.89(dd, J=6.0,13.4Hz,1H),3.81(dd,J=6.0, 8.5Hz,1H),6.55(d,J=15.5Hz,1H)	2.78(dd,J=8.6,13.4Hz,1H),2.91(dd,J=6 .0,13.4Hz,1H),3.92(ABq,J=13.5Hz,1H) ,3.90(m,1H),9.01(s,1H),10.78(s,1H)	I
IR (\(\nu\) cm ⁻¹) (KBr)	3700-2400(br),3312, 1629,1329,1144	3700-2200(br),1670, 1318,1152	I
mp (decomp.)	138-139	02-69	l
*	22	22	æ
R 18		CH2-	-NH.
R 1	CH2-	CH2-	IZ CH
Example No.	7.7	7.8	6 2
	R^{-1} R^{-18} $*$ $mp (decomp.)$ $IR (\nu cm^{-1})$ (KBr)	R 1 R (ν cm ⁻¹) (KBr) (C) (KBr) (A29,1329,1144	R 1 mp (decomp.) IR (ν cm ⁻¹) (KBr) CH ₂ - CH ₂ - CH ₂ - CH ₂ - R 138-139 3700-2400(br),3312, 1629,1329,1144

Он (Ia)	'H-NMR([§] ppm) d ₆ -DMSO	0.89(1,J=6.7Hz,3H),2.62(1,J=7.6Hz,2H),2.96(d d,J=7.0,13.9Hz,1H),3.10(dd,J=5.4,13.9Hz,1H) ,4.19(d1,J=6.9,8.2Hz,1H),5.30(d,J=8.2Hz,1H),	0.88(t,J=6.9Hz,3H),2.55-2.73(m,2H),2.9 7(dd,J=8.4,13.8Hz,1H),3.24(dd,J=4.8,13. 8Hz,1H),4.35(m,1H),4.98(m,1H) (CDCb)	0.84(t,J=7.1Hz,3H),2.57-2.70(m,2H),2.97(d d,J=8.4,13.9Hz,1H),3.25(dd,J=4.8,13.9Hz,1 H),4.35(m,1H),4.96(d,J=9.6Hz,1H) (CDCl ₃)	2.41(s,3H),3.01(dd,J=6.0,14.4Hz,1H),3. 12(dd,J=4.5,14.4Hz,1H),3.67(t,J=5.4Hz, 1H),6.79(m,1H),6.89(m,1H),10.59(s,1H)	3.03(dd,J=6.5,15.1Hz,1H),3.15 (dd,J=4.7,14.1Hz,1H),3.64(t, J=5.1Hz,1H),10.68(s,1H)	2.81 (dd,J=9.2,13.7Hz,1H),3.03(dd,J=5.4,13.7H z,1H),3.94(dt,J=5.4,9.2Hz,1H),6.66(d,J=15.2H z,1H),7.16(d,J=15.2Hz,1H),8.01(d,J=9.2Hz,1H)	2.81(dd,J=9.2,13.7Hz,1H),3.00(dd,J =5.6,13.7Hz,1H),4.01(ABq,J=13.7Hz ,2H),4.01(m,1H),7.65(d,J=8.3Hz,1H)	0.90-1.68(m,9H),1.78(m,1H),2.74 (m,1H),3.00-3.20(m,2H),3.77(m, 1H)6.45(br.s,1H),6.77(br.s,1H)
	IR (v cm ⁻¹) (KBr)	2300-3700br,3426,3318, 1713,1330,1159	2400-3600br,3340,1736, 1334,1142(CHCl _b)	2300-3700br,3240, 1725,1341,1144	3421,1580,1333, 1421,1153	3413,1594,1456, 1416,1157	2400-3700br,3252,1765, 1725,1301,1140	2200-3700br,3268,1726, 1321,1152(film)	3413,2931,1720,1585, 1455,1421,1313,1144
R¹ R¹®·SO₂NH * COOH	mp (decomp.) (C)	121-122	ië	06-68	>250	foam	1	ı	l
æ.	*	22	22	æ	R	R	R	22	R
	R 18	CH ₃ (CH ₂)4	CH ₃ (CH ₂) ₇ —	CH ₃ (CH ₂) ₃ —	CI CH ₃	O O		-ZH2-CH2-	-NF
	R 1	-2H2-	-2H2-	-ZH2-CH2-	TZ GH	CH ₂ .	CH2-	CH ₂ -	CH ₂ .
	Example No.	7.1	7.2	7.3	7.4	9 2	2.2	7.8	6 2

Table 19

	Elemental analysis			C ₂₄ H ₁₉ N ₃ O ₅ S•1.3H ₂ O Calc. C:59.45 H:4.49 N:8.67 S:6.61 Foun.C:59.43 H:4.45 N:8.59 S:6.58		-
æ	IR (\(\nu\) cm\\) (KBr)	1704,1596 1349,1164	1576,1356 1139	1732,1342 1167	1745,1590 1316,1157	1594,1456 1200,1188
Р ¹ R ¹⁸ -SO ₂ NH <mark>→</mark> СООН (Ia)	mp (decomp.)	153-155	>130	128-130	210-214	198-200
) ₂ NH	*	R	R	R	R	R
R ¹⁸ SO ₂ I	*- X	-{_}\\n _{B₁}	n-C ₈ H ₁₇ -			
	. R	CH ₂	TX TX	N CH2.	The CH2.	H CH ₂ .
	Example No.	8 0	8 1	8 2	8 3	8 4

Table 20

	¹ H-NMR(ô ppm) d ₆ -DMSO	2.65(dd,J=8.9,13.6Hz,1H), 2.82(dd, J=6.6,13.6Hz,1H),3.86(m,1H),7.75 (d,J=7.8Hz,2H),7.87(d,J=8.7Hz,2H)	2.62(dd,J=8.6,13.5Hz,1H),2.81(dd,J=6. 5,13.6Hz,1H),3.09(s,6H),3.83(m,1H),6 .86(d,J=9.0Hz,2H),7.83(d,J=8.8Hz,2H)	3700-2400(br),3357,1686, H),3.76(m,1H),8.02(d,J=8.7Hz,1H),8.80(s,1H),8.8 (m,1H),8.02(d,J=8.7Hz,1H),8.80(s,1H),8.8 (d,J=1.7Hz,1H),9.06(s,1H),10.59(d,J=1.7Hz,1H)
HOH (lb)		3700-2400br,3273, 1633,1338,1166	3700-2400br,2921, 1672,1314,1165,	3700-2400(br),3357,1686, 1641,1314,1155
R¹ ⊢ R¹®SO₂NH ★CONHOH	mp (decomp.)	157-160	138-142	206-207
R ¹⁸ .9	*	R	22	တ
	R 18	N-N-N R	MezN -N:N - R	S CHANGE
	R.	CH2-CH2-	-zH2-CH2-	-2H2-
	Example No.	8 5	9 8	2.8

Table 21

		T		
	H-NMR(& ppm) de-DMSO	2.75(dd,J=9.1,13.7Hz,1H),2.98(dd, J=5.5,13.7Hz,1H),3.96(ddd,J=5.5, 9.1,9.1Hz,1H),8.51(d,J+9.1Hz,1H)	2.74(dd,J=9.1,13.6Hz,1H),2.96(dd,J =5.7,13.6Hz,1H),3.09(s,6H),3.93(dt, J=5.7,9.1Hz,1H),8.39(d,J=9.1Hz,1H)	2.71(dd,J=9.1,13.7Hz,1H),2.93(dd,J=5.6 ,13.7Hz,1H),3.84(dt,J=5.6,9.1Hz,1H),8. 11(d,J=9.1Hz,1H),8.78(s,1H),9.06(s,1H)
ООН (Ia)	IR (\(\nu\) cm ⁻¹) (KBr)	2400-3600br,3426,3296, 1698,1350,1167	2200-3700br,3431, 1735,1391,1154	2300-3700br,3358, 3262,1718,1686, 1660,1313,1159
^{R¹} Р¹®-SO ₂ NH <mark>*,</mark> СООН (Ia)	mp (decomp.)	172-174	93-94	203-204
<u>я</u>	*	R	R	S
	R 18	R-N-N-N-R	MezN \\ \rightarrow N:N \\ R	S
	R.	CH₂-	-2H2-	CH2-
	Example No.	8 5	9 8	8.7

Table 22

			-,		
	Elemental analysis	1	C ₁₇ H ₂₀ N ₂ O ₆ S ₂ •0.9Ethylether Calc. C:51.63 H:6.10 N:5.85 S:13.38 Foun.C:51.23 H:6.17 N:5.87 S:13.11	C ₁₈ H ₂₁ N ₃ O ₆ S ₂ •0.8Ethylether Calc. C:51.05 H:5.86 N:8.42 S:12.86 Foun.C:50.75 H:5.89 N:8.15 S:12.47	C ₂₁ H ₁₉ BrN ₂ O ₆ S ₂ •0.5CF ₃ COOH Calc. C:44.30 H:3.30 Br:13.40 N:4.70 S:10.75 Foun.C:44.62 H:3.52 Br:13.07 N:4.64 S:10.85
(la)	IR (\(\nu\) cm ⁻¹) (KBr)	1719,1390 1229	1734,1461 1327,1158	1724,1325 1168	1735,1598 1327,1185
R¹ ⁸ ·SO ₂ NH [★] ,COOH (Ia)	mp (decomp.)	103-106	66-96	110-112	98-101
SO ₂ N	*	R	ж	ж	R
R. 8	R ! 8	-S-N()	-N-S-V	-C-N-N-S-W-N-S-	Br \S-N-\
	R.	CH ₂ .	(СН ₃)2СН-	(CH ₃) ₂ CH-	CH2-CH2-
	Example No.	8 8	68	0 6	9.1

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Example 92 (Method B)

Process 1

To a solution of D-valine methylester hydrochloride (XV-2) (755 mg, 4.5 mmol) in dichloromethane(12 ml) was added N-methylmorpholine (1.49 ml, 3×4.5 mmol) and 5-bromo-2-thiophensulfonyl chloride (1.24 g, 1.05×4.5 mmol) was added under ice-cooling. After being stirred for 15 h at room temperature, the reaction mixture was washed with 2N HCl, 5% NaHCO₃, and water. The organic layer was concentrated in vacuo, and dried over Na₂SO₄. The residue was subjected to silica gel column chromatography and the fractions eluting with ethyl acetate / hexane = 1/3 were collected and washed with n-hexane to give 1.32 g of the desired compound (XVII-1). Yield 82 %. mp. 109-110°C.

Elemental analysis C₁₀H₁₄BrNO₄S₂

Calcd. : C; 33.71 H; 3.96 Br; 22.43 N; 3.93 S;1 8.00

Found: C; 33.75 H; 3.89 Br; 22.43 N; 3.96 S; 17.86

 $[\alpha]_D: -34.5 \pm 0.7 (c=1.012 \text{ CHCl}_3 25^{\circ}C)$

 $IR(CHCl_3, \nu \text{ max cm}^{-1})1737,1356,1164,1138$

NMR (CDCl₃, δ ppm): 0.89(d, J=6.8 Hz, 3H), 1.00(d, J=6.8 Hz, 3H), 2.00 (m, 1H), 3.60(s, 3H), 3.83(dd, J=5.2, 10.0 Hz, 1H), 5.20(d, J=10.0 Hz, 1H), 7.04(d, J=4.1 Hz, 1H), 7.32(d, J=6.8 Hz, 3H), 3.83(dd, J=5.2, 10.0 Hz, 1H), 5.20(d, J=10.0 Hz, 1H), 7.04(d, J=4.1 Hz, 1H), 7.32(d, J=6.8 Hz, 3H), 3.83(dd, J=5.2, 10.0 Hz, 1H), 5.20(d, J=10.0 Hz, 1H), 7.04(d, J=4.1 Hz, 1H), 7.32(d, J=6.8 Hz, 3H), 3.83(dd, J=5.2, 10.0 Hz, 1H), 5.20(d, J=10.0 Hz, 1H), 7.04(d, J=4.1 Hz, 1H), 7.32(d, J=6.8 Hz, 3H), 3.83(dd, J=5.2, 10.0 Hz, 1H), 5.20(d, J=10.0 Hz, 1H), 7.04(d, J=4.1 Hz, 1H), 7.32(d, J=6.8 Hz, 3H), 3.83(dd, J=5.2, 10.0 Hz, 1H), 7.32(d, J=6.8 Hz, 3H), 3.83(dd, J=6.8 Hz, 3H), 3.83(

J=4.1 Hz, 1H

Process 2

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To a degassed solution of 400 mg (1.12 mmol) of compound (XVII-1) in 5 ml of dimethylformamide was added 222 mg (1.5 x 1.12 mmol) of 4-methoxyphenylacetylene and 21 mg(0.1 x 1.12 mmol) of copper iodide (I) under an argon atmosphere. Then 39 mg (0.05 x 1.12 mmol) of bis(triphenylphosphine)palladium dichloride (II) and 0.47 ml (3 x 1.12 mmol) of triethylamine were added to the reaction mixture. The resulting mixture was degassed and stirred overnight under an argon atmosphere at 50 °C. The reaction mixture was diluted with ethyl acetate. The organic later was washed with 1N HCl, 5 % NaHCO₃, and water, dried over Na₂SO₄, and concentrated in vacuo. The resulting residue was column chromatographed on silica gel. The fractions eluting with n-hexane / ethyl acetate = 2/1 were collected and recrystallized from ethyl acetate / n-hexane to give 392 mg of the desired compound (XVIII-1). Yield 86 %. mp. 131-132°C.

15 Elemental analysis C₁₉H₂₁NO₅S₂·0.2 H₂O

Calcd. : C; 55.51 H; 5.25 N; 3.41 S; 15.60

Found: C; 55.80 H; 5.19 N; 3.38 S; 15.36

 $IR(KBr, \nu \ \max \ cm^{-1}): 3268, 2203, 1736, 1604, 1524, 1348, 1164.$

NMR(CDCl₃, δ ppm): 0.90(d, J=6.6 Hz, 3H), 1.00(d, J=7.0 Hz, 3H), 2.00(m, 1H), 3.60(s, 3H), 3.84(s, 3H), 3.86(dd, J=5.0, 10.2 Hz, 1H), 5.21(d, J=10.2 Hz, 1H), 6.90(d, J=9.0 Hz, 2H), 7.44(d, J=9.0 Hz, 2H), 7.12(d, J=4.0 Hz, 1H), 7.44(d, J=4.0 Hz, 1H).

Process 3

To a solution of 407 mg (1 mmol) of compound (XVII-1) in 8 ml of tetrahydrofuran and 8 ml of methanol was added 5.1 ml of 1N NaOH. The resulting mixture was stirred for 6 h at 60 °C. The reaction mixture was concentrated in vacuo to remove an organic solvent, and the residue was diluted with ethyl acetate. The mixture was acidified with aqueous solution of citric acid and extracted with ethyl acetate. The organic layer was washed with brine, dried over Na₂SO₄, and concentrated in vacuo to give 373 mg of compound (Ia-2-1). Yield 100%. mp. 147-

148℃.

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IR (KBr, ν max cm⁻¹): 1710,1604,1351,1216.

Elemental analysis $C_{18}H_{19}NO_5S_2 \cdot 0.2H_2O$

Calcd. : C; 54.45 H; 4.92 N; 3.53 S; 16.15

Found: C; 54.39 H; 4.93 N; 3.79 S; 15.96

Example 93 - 156

The compounds which were shown in Tables 23 to 30 were synthesized in a manner similar to those described in Example 92.

Table 23

	Elemental analysis	I	C ₂₆ H ₂₂ N ₂ O ₅ S Calc. C:65.81 H:4.67 N:5.90 S:6.76 Foun.C:65.34 H:4.90 N:5.56 S:6.40	1	I	1	I	C ₂₆ H ₂₀ N ₂ O ₆ S-0.4H ₂ O Calc. C:63.00 H:4.23 N:5.65 S:6.47 Foun.C:62.99 H:4.32 N:5.82 S:6.76	C ₂₅ H ₂₁ N ₃ O ₄ S•0.8H ₂ O Calc. C:63.36 H:4.81 N:8.87 S:6.77 Foun.C:63.45 H:4.92 N:8.77 S:6.57
(la)	IR (\(\nu\) cm ⁻¹) (KBr)	1590,1316 1137	1747,1323 1134	1724,1325 1135	1739,1336 1163	1710,1511 1329,1161	1725,1618 1373,1163	1706,1606 1350,1164	1735,1633 1321,1173
R¹8-SO₂NH ★ COOH (8	mp (decomp.) (C)	165-170	223-226	216-218	111-114	178-180	105-108	>250	176-177
O ₂ NH Č	*	Ж	R	R	R	R	R	R	R
R ¹⁸ .SO ₂	R 18	-{_}o≡o-{_}	H₃co-⟨>-C≣c-⟨}	но-{_}с≡с-{_}-	H₃coco-{}-C≡C-{}	F-\\-CEC-_\-	O ₂ N-{}-C≡C-{}-	-{_}С≣С-{_}-ОООН	H ₂ N ← C≡C ← F
	R 1	CH_{2}	M CH ₂ .	$\text{CH}_{2^{-}}$	H CH2-	(CH2-	M CH ₂ .	CH ₂ ·	FX G.
	Example No.	9.3	9.4	9 5	9 6	9.7	8 6	6 6	100

Table 24

R¹ R¹8-SO₂NH ★ COOH (la)	Elemental analysis	C ₂₆ H ₂₂ N ₂ O ₄ S•0.2H ₂ O Calc. C:67.57 H:4.89 N:6.06 S:6.94 Foun.C:67.66 H:4.77 N:6.09 S:6.71	-	l	!	1	C ₁₉ H ₁₈ N ₂ O ₆ S•0.1H ₂ O Calc. C:56.46 H:4.54 N:6.93 S:7.93 Foun.C:56.30 H:4.37 N:7.14 S:7.85	l	I
	IR (\(\nu\) cm ⁻¹) (KBr)	1736,1618 1398,1168	1735,1654 1399,1164	1732,1631 1372,1148	1600,1558 1336,1171	1795,1718 1331,1166	1719,1595 1344,1167	1728,1631 1372,1148	1728,1332 1172
	mp (decomp.) (C)	227-229	230-233	234-236	>200 decomp.	146-149	231-232	166-169	163-165
O ₂ NH	*	R	R	R	R	R	R	R	Ж
R ¹⁸ SO	R ¹⁸	H₃C-{}-C≣C-{}-	нс≡с-{_}-с≡с-{_}-	Me ₂ N-⟨⟩-C≣C-⟨⟩-	H_3CO $C = C$	H ₃ CO-⟨>-C≣C-⟨}>-	O_2N \subset \subset \subset \subset \subset \subset \subset \subset	H ₂ N-{}C≣C-{}	но-{_}с≡с-{_}
	R ¹	M CH ₂ -	M CH ₂ .	CH ₂ .	CH ₂ -	(СН ₃)2СН-	(СН ₃)2СН-	(CH ₃) ₂ CH-	(СН3)2СН-
	Example No.	101	102	103	104	105	106	107	108

Table 25

Elemental analysis	I		C ₂₁ H ₂₃ NO ₅ S+1.3H ₂ O Calc. C:59.36 H:6.07 N:3.30 S:7.55 Foun.C:59.36 H:6.06 N:3.50 S:7.44	1		1	C ₂₃ H ₁₈ FNO ₄ S•0.3H ₂ O Calc. C:64.41 H:4.37 F:4.43 N:3.27 S:7.48 Foun.C:64.37 H:4.38 F:4.96 N:3.31 S:7.24	1
IR (\(\nu\) cm ⁻¹) (KBr)	1720,1656 1319,1165	1724,1635 1366,1158	1711,1683 1600,1328 1159	1732,1680 1329,1167	1735,1651 1348,1165	1727,1604 1335,1182	1725,1663 1399,1197	1728,1332 1172
mp (decomp.) (C)	187-189	111-114	161-162	157-159	133-136	183-185	166-168	163-165
*	æ	æ	Я	R	24	24	24	R
R ¹⁸	H ₃ C-{}-C≡C-{}}-	-{_}c≡c-{_}-	H3CO-{_}C≣C-{_}	-{_}c≘c-{_}}-os€H	-{_}o≘o-{_}oo¢н	-{_}⊃≣O-{_}>⊃ [¢] H	-{_}-0≣0{_}-4	но-{_}-с≡с-{_}-
R	(СН3)2СН-	(CH ₃) ₂ CH-	(CH ₃) ₃ C-	СН ₃ СН ₂ (СН ₃)СН-	CH ₂ -	CH ₂ -	CH2-	(CH ₃) ₂ CH-
Example No.	109	1 1 0	1 1 1	112	113	114	115	116

Table 26

R¹®-SO₂NH ♣COOH (la)	Elemental analysis		I	1	-	l	l	-	C ₁₈ H ₁₉ NO ₅ S ₂ -0.2H ₂ O Calc. C:54.45 H:4.92 N:3.53 S:16.15 Foun.C:54.39 H:4.93 N:3.79 S:15.96
	IR (\(\nu\) cm ⁻¹) (KBr)	1720,1656 1319,1165	1724,1635 1366,1158	1585,1318 1153	1605,1523 1340,1151	1604,1524 1336,1173	1721,1620 1339,1163	1729,1675 1340,1168	1710,1604 1351,1216
	mp (decomp.) (C)	187-189	111-114	167-169	l	I	103-106	180-182	147-148
_ \	*	æ	R	R	R	R	ж	R	X
H ^{IB.} SO.	R 18	H3C-{_}-C≡C-{_}	F-CEC-C	~ S⊃-c≡c-	C≡C-CsC-Cs NO ₂	H3CO-{\rightarrow}-C≣C-{\rightarrow}-	F-CEC-S	H ₃ C ← C≡C ← S	H₃CO-{}C≘C-{}}
	R.	(CH ₃) ₂ CH-	TX TX	TX TX	CH2-CH2-	TX HO	CH2.	CH2.	(CH ₃) ₂ CH-
	Example No.	117	1 1 8	1 1 9	120	121	122	123	124

Table 27

R¹ R¹® SO₂NH ★ COOH (Ia)	Elemental analysis	C ₁₆ H ₁₉ NO ₄ S ₂ -0.2H ₂ O Calc. C:56.73 H:5.13 N:3.68 S:16.83 Foun.C:57.03 H:5.30 N:3.89 S:16.56		C ₂₂ H ₁₉ NO ₅ S _{2*} 0.2H ₂ O Calc. C:59.36 H:4.39 N:3.15 S:14.41 Foun.C:59.43 H:4.61 N:3.25 S:14.02	_	C ₂₁ H ₁₆ FNO ₄ S ₂ Calc. C:58.73 H:3.75 F:4.42 N:3.26 S:14.93 Foun.C:58.66 H:3.93 F:4.52 N:3.33 S:14.41		-	I
	IR (v cm ⁻¹) (KBr)	1712,1350 1163	1710,1499 1356,1165	1695,1334 1184	1710,1329 1180	1734,1699 1324,1105		1	I
	mp (decomp.) (C)	157-158	154-156	149-150	161-164	155-158	ł	-	-
PNS(NH)	*	R	R	R	æ	R	R	R	R
R ¹⁸ .SO ₂	R 18	H3C-{_S-C≡C-_S+	F-CEC-(S)	H ₃ CO-⟨}-C≡C-⟨_S	H3C-{\}_C≡C-{\}_S	F-C=C-CS	6400 6400 6400	H³co H³co H³co	CEC NO2
	. A	(СН3)2СН-	(CH ₃) ₂ CH-	CH2-	CH ₂ -	CH ₂ -	-CH2-	-CH2-	CH ₂ -
	Example No.	125	126	127	128	1 2 9	130	131	132

Table 28

R ¹ R ¹⁸ ·SO ₂ NH * COOH (Ia)	Elemental analysis	I	-			I	Ţ	1	I
	IR (\(\nu\) cm ⁻¹) (KBr)	I	I	ļ	ŀ	l	ł.	ţ	I
	mp (decomp.) (C)	l	l	l	l	-	l	1	1
_ _HN2C	*	Ж	R	R	R	R	R	ಜ	æ
R ¹⁸ -SO.	R ' 8	$\left\langle \begin{array}{c} \\ \\ \\ \\ \end{array} \right\rangle = \mathbb{C} = \mathbb{C} - \left\langle \begin{array}{c} \\ \\ \\ \\ \end{array} \right\rangle$	CH ₃ (CH ₂) ₅ -CEC	-{_}5=5-{_}ооын	~\s_2=\g\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	F-CEC-CS	Br CEC S	CI-CEC-CS	HO—C≣C—CS—OH
	. R	CH2-CH2-	CH2-	CH ₂ -	CH2-	CH2-	CH ₂ -	CH2-	СН₂-
	Example No.	133	134	135	136	137	138	139	140

Table 29

R¹ ⁸ -SO₂NH ★ COOH (Ia)	Elemental analysis	1		1		-	1	1	-
	IR (\(\nu\) cm ⁻¹) (KBr)	l	-	-	l	ı	-	1	l
	mp (decomp.) (C)	I	l	1		ļ			1
N ₂ NH	*	R	R	R	Ж	23	R	24	24
R ¹⁸ ·SO,	R 18	-(\$)-0≣0-(\$)-		F_3C $\left(-\right)$ C $\equiv C$ $\left(-\right)$ C		Meoc-C=C-Cs	-CEC-CS	H00C \\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	Meooc C=c S
	R.	CH2-	CH2-	-CH ₂ -	-CH ₂ -	-CH2-	-ZH2CH2-	-CH2-	CH ₂ -
	Example No.	141	142	143	144	145	146	147	148

Table 30

	Elemental analysis	1		-	1	ı	-	1	-
R¹ R¹8-SO₂NH ★СООН (Ia)	IR (\(\nu\) cm ⁻¹) (KBr)	l	l	-	ı	ı	1	ı	l
	mp (decomp.)	l	1	i	ı	I	l	ı	I
O ₂ NH´	*	ж	æ	R	R	~	ĸ	æ	æ
P ¹⁸ -SO	R 18	H ₂ NOC CEC S	-√s -с≡с-√->-оно	$-\sqrt{s} - c = c - \sqrt{s}$	H ₂ N-CEC-(S)	Me ₂ N CEC S	MeO ₂ S - CEC - S	HS-CEC-S	NC CEC S
	R.	CH ₂ -	CH2-	CH2-	-CH2-	-CH2-	-CH2-	-cH2-	-CH2-
	Example No.	149	150	151	152	153	154	155	156

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Example 157, 158

MeO So₂-N COOMe Process 1

XVIII-2

MeO So₂-N COOH

$$I_{A}$$
-2-66, I_{A} -2-67

Process 1 ($R^2 = CH_3$)

To a solution of 150 mg (0.33 mmol) of compound (XVIII-2) in 2 ml of dimethylformamide which was synthesized the same manner as those described in Example 96 was added 227 mg (5 x 0.33 mmol) of potassium carbonate and 0.1 ml (5 x 0.33 mmol) of methyl iodide, and the resulting mixture was stirred overnight at room temperature. The reaction mixture was poured into water and extracted with ethyl acetate. The organic layer was washed with water, dried over Na₂SO₄, and concentrated in vacuo to give 373 mg of N-methyl derivative as an oil. Yield 91%.

Elemental analysis C24H23NO5S2

Calcd. : C; 61.39 H; 4.94 N; 2.98 S; 13.66

Found: C; 61.22 H; 5.18 N; 2.93 S; 13.27

Further, a solution of 140 mg of the above oily compound which was obtained the above process in 2 ml of methanol was added 0.6 ml of 1N NaOH, and the resulting mixture was stirred overnight at room temperature. The reaction mixture was acidified with 2N HCl and extracted with ethyl acetate. The organic layer was washed with water, dried over Na₂SO₄, and concentrated in vacuo to give 105 mg of compound (Ia-2-66) (R= Me). Yield 77 %. mp. 185 - 186°C.

20 Elemental analysis C23H21NO5S

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Calcd. : C; 60.64 H; 4.65 N; 3.07 S; 14.08

Found: C; 60.56 H; 4.84 N; 3.01 S; 13.94.

IR (KBr. v max cm⁻¹): 3600-2300br, 3426, 2203, 1710, 1604, 1503, 1344, 1151.

NMR (d₆-DMSO, δ ppm): 2.88(s, 3H), 2.93(dd, J=12.0, 10.2 Hz, 1H), 3.19 (dd, J=14.2, 5.6 Hz, 1H), 3.81(s, 3H), 4.74(dd, J=5.4, 10.2 Hz, 1H), 6.99-7.04(m, 2H), 7.20-7.35(m, 7H), 7.52-7.56(m, 2H), 6.90(d, J=9.0 Hz, 2H), 7.44(d, J=9.0 Hz, 2H), 7.12(d, J=4.0 Hz, 1H), 7.44(d, J=4.0 Hz, 1H).

The compound (Ia-2-67) ($R^2 = CH_2Ph$) was synthesized in the same manner as those described in Example 157,.

 $IR(KBr, v max cm^{-1}): 2200,1722,1340,1151.$

NMR (d₆-DMSO, δ ppm) : 2.94(dd, J=7.6, 13.8 Hz, 1H), 3.19(dd, J=7.2, 14.4 Hz, 1H), 3.83(s, 3H), 4.29(d, J=16.2 Hz, 1H), 4.62(d, J=16.2 Hz, 1H) (Only characteristic peaks are shown.)

Example 159 (Method C)

Process 1

To a solution of 500 mg (1.4 mmol) of compound(XVII-2) which was obtained Example 96 in 12 ml of dry tetrahydrofuran was added 387 mg (2 x 1.4 mmol) of powdery potassium carbonate, 319 mg (1.5x1.4 mmol) of 4-methoxyphenylboronic acid and 81 mg (0.05 x 1.4 mmol) of tetrakis(triphenylphosphine)palladium. The resulting mixture was stirred under argon atmosphere for 48 h at 75°C. The reaction mixture was diluted with ethyl acetate. The organic layer was washed with 1N HCl, 5% NaHCO₃ aq., and water, dried over Na₂SO₄, and concentrated in vacuo. The residue

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was column chromatographed on silica gel. The fractions eluting with n-hexane / ethyl acetate = 3/1 were collected and recrystallized from n-hexane to give 447 mg of the desired compound (XIX-1). Yield 83 %. mp. 122-123℃.

Elemental analysis C₁₇H₂₁NO₅S₂

Calcd. : C; 53.25 H; 5.52 N; 3.65 S; 16.72

Found: C; 53.26 H; 5.50 N; 3.69 S; 16.63

 $[\alpha]_D$ -21.7 ± 0.6 (c=1.000 DMSO 25°C)

IR (KBr, $v \max cm^{-1}$): 1735,1605,1505,1350,1167,1136

NMR (CDCl₃, δ ppm): 0.90(d, J=7.0 Hz, 3H), 1.00(d, J=6.6 Hz, 3H), 2.10(m, 1H), 3.54(s, 3H), 3.85(s, 3H), 3.87(dd, J=5.0, 10.2 Hz, 1H), 5.20(d, J=10.2 Hz, 1H), 6.94(d, J=9.0 Hz, 2H), 7.52(d, J=9.0 Hz, 2H), 7.11(d, J=4.0 Hz, 1H), 7.49(d, J=4.0 Hz, 1H).

Process 2

To a solution of 390 mg (1.01 mmol) of compound (XIX-1) in 8ml of tetrahydrofuran and 8ml of methanol was added 5.1 ml of 1N NaOH, and resulting mixture was stirred at 60℃ for 6 h. The reaction mixture was concentrated in vacuo to remove an organic solvent. The resulting residue was diluted with ethyl acetate. The mixture was acidified with aqueous solution of citric acid and extracted with ethyl acetate. The organic layer was washed with brine, dried over Na₂SO₄, and concentrated in vacuo to give 373 mg of compound (Ia-3-1). Yield 100%. mp. : 174-

20 176℃

 $IR(KBr, v max cm^{-1}): 1735, 1503, 1343, 1163.$

Example 160 - 175

The compounds which were shown in Tables 31 to 32 were synthesized in a manner similar to those described in Example 159,.

Table 31

	Elemental analysis			-	C ₂₂ H ₂₀ N ₂ O ₄ S ₃ •0.4H ₂ O Calc. C:55.07 H:4.37 N:5.84 S:20.05 Foun.C:55.35 H:4.43 N:6.04 S:19.65	ľ	_	C ₁₅ H ₁₆ FNO ₄ S ₂ -0.1H ₂ O Calc. C:50.15 H:4.55 F:5.29 N:3.90 S:17.85 Foun.C:49.99 H:4.58 F:5.22 N:4.05 S:17.77	C ₁₆ H ₁₉ NO ₄ S ₃ Calc. C:49.85 H:4.97 N:3.63 S:24.95 Foun.C:49.70 H:5.00 N:3.93 S:24.96
(E	IR (v cm ⁻¹) (KBr)	1667,1337 1180	1670,1339 1194	1725,1598 1371,1185	1735,1341 1159	1735,1503 1343,1163	1713,1353 1163	1702,1504 1352,1168	1747,1324 1159
R¹ R¹8-SO₂NH ★ COOH (Ia)	mp (decomp.) (C)	93-96	157-159	168-171	226-230	174-176	165-167	146-147	157-159
L NN2C	*	В	R	В	R	R	R	24	ద
R ¹⁸ -S(R.1.8	H ₃ CO-\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	H3C \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\		H ₃ CS	H3CO-{_}\C	H3C	\S\-\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	H ₃ CS \\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\
	R.1	HZ HZ	TZ HÖ	IZ TO	CH ₂ .	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-
	Example No.	160	161	162	163	164	165	166	167

Table 32

		No. 18	L SINIT	R. 502NH * COOH (12)		
Example No.	۳.	R -8	*	mp (decomp.) (C)	IR (\(\nu\) cm ⁻¹) (KBr)	Elemental analysis
168	CH2-CH2-	H ₃ CO-{}	24	161-165	1735,1698 1374,1163	C ₂₀ H ₁₉ NO ₅ S ₂ Calc. C:57.54 H:4.59 N:3.35 S:15.36 Foun.C:57.62 H:4.72 N:3.52 S:15.27
169	-CH2-	H ₃ C-{}-	~	166-167	1713,1609 1378,1194	G ₂₀ H ₁₉ NO ₄ S ₂ Calc. C:59.83 H:4.77 N:3.49 S:15.97 Foun.C:59.77 H:4.86 N:3.61 S:15.86
170	CH2-CH2-	F S	24	174-175	1721,1654 1365,1148	C ₁₉ H ₁₆ FNO ₄ S ₂ Calc. C:56.28 H:3.98 F:4.09 N:3.45 S:15.82 Foun.C:56.33 H:4.09 F:4.65 N:3.65 S:15.84
171	CH ₂ -	H ₃ CS	ಜ	203-205	1750,1730 1428,1325 1155	C ₂₀ H ₁₉ NO ₄ S ₃ ·0.2H ₂ O Calc. C:54,95 H:4.47 N:3.20 S:22.00 Foun.C:55.05 H:4.52 N:3.34 S:22.04
172	CH ₂ -	H ₂ N ₄ H	æ	l	i	I
173	CH ₂ -	Me ₂ N-S	8	! -		l
174	-CH2-	F ₃ C	24	Į	l	!
175	-ZHD-CH2-	NC S	&	ı	1	I

Example 176 (Method D)

HCl
$$H_2N$$
 COO^tBu $Process 1$ O_2N SO_2-N COO^tBu Y $XX-1$ $XX-1$

Process 1

5

10

To a solution of 10 g (47.68 mmol) of D-valine tert-butyl ester hydrochloride (XV-3) in 100 ml of dichloromethane was added 15.7 ml (3 x 47.68 mmol) of N-methylmorpholine and 14.1 g(1.2 x 47.68 mmol) of 4-nitrobenzenesulfonyl chloride under ice-cooling. After being stirred for 5 h at room temperature the reaction mixture was washed with 2N HCl, 5% NaHCO₃, water. The organic layer was dried over Na₂SO₄ and concentrated in vacuo, and the resulting residue was recrystallized from dichloromethane / n-hexane to give 13.3g of the desired compound (XX-1). Yield 77.8%. mp. 89-90°C.

Elemental analysis C₁₅H₂₂N₂O₆S

Calcd. : C; 50.27 H; 6.19 N; 7.82 S; 8.95

Found: C; 50.04 H; 6.10 N; 7.89 S; 8.84

15 [α]_D -2.9 \pm 0.8(c=0.512 DMSO 23 $^{\circ}$ C)

IR(KBr, v max cm⁻¹): 3430br, 3301, 1722, 1698, 1525, 1362, 1348, 1181, 1174, 1159.

Process 2

A solution of 13.29 g (37.08 mmol) of compound (XX-1) in 200 ml of methanol was hydrogenated using 10% Pd/C (1g) for 2h at room temperature. The reaction mixture was filtered off and the filtrate was concentrated in vacuo. The residue was recrystallized from acetone / n-hexane to give 11.5g of amine derivative (XXI-1). Yield 94.4%. mp. 164-166℃

Elemental analysisC₁₅H₂₄N₂O₄S

Calcd. : C; 54.86 H; 7.37 N; 8.53 S; 9.76

Found: C; 54.84 H; 7.33 N; 8 63 S; 9.50

10 [α]_D +10.3±1.0(c=0.515 DMSO 23°C)

IR(KBr, v max cm⁻¹): 3461, 3375, 1716, 1638, 1598, 1344, 1313.

NMR(d-DMSO, δ ppm) : 0.80(d, J=6.8 Hz, 3H), 0.82(d, J=6.6 Hz, 3H), 1.23(s, 9H), 1.83(m, 1H), 3.30(m, 1H), 5.86(s, 2H), 6.56(d, J=8.8 Hz, 2H), 7.36(d, J=8.6 Hz, 2H), 7.47(d, J=9.6 Hz, 1H)

15 Process 3

20

25

To a solution of 328 mg (1mmol) of compound (XXI-1) in 10 ml of dichloromethane was added 0.33 ml (3 x 1 mmol) of N-methylmorpholine and 280 mg (1.5 x 1 mmol) of 4-(methylthio)benzoyl chloride under ice-cooling. The reaction mixture was stirred overnight at room temperature. To the reaction mixture was added ethyl ether and precipitation were collected and washed with ice-water and ethyl ether, The solid were recrystallized from acetone / ethyl ether to give 433 mg of the desired compound (XXII-1). Yield 90.5%. mp. 235-238°C.

Elemental analysisC23H30N2O5S2

Calcd. : C; 57.72 H; 6.32 N; 5.85 S; 13.40

Found: C; 57.63 H; 6.28 N; 5.86 S; 13.20

 $[\alpha]_D +5.7 \pm 0.9 (c=0.512 DMSO 25^{\circ})$

 $IR(KBr, v \max cm^{-1}): 3366, 3284, 1713, 1667, 1592, 1514, 1498, 1341, 1317.$

 $NMR(d_6\text{-DMSO}, \ \delta \ ppm) \ : \ 0.82(d, \ J=6.6 \ Hz, \ 3H), \ 0.84(d, \ J=6.8 \ Hz, \ 3H), \ 1.22(s, \ 9H), \\ 1.91(m, \ 1H), \ 2.55(s, \ 3H), \ 3.32(s, \ 3H), \ 3.44(dd, \ J=6.2, \ 8.6 \ Hz, \ 1H), \ 7.40(d, \ J=8.6 \ Hz, \ 2H), \\ 1.91(m, \ 1H), \ 2.55(s, \ 3H), \ 3.32(s, \ 3H), \ 3.44(dd, \ J=6.2, \ 8.6 \ Hz, \ 1H), \ 7.40(d, \ J=8.6 \ Hz, \ 2H), \\ 1.91(m, \ 1H), \ 2.55(s, \ 3H), \ 3.32(s, \ 3H), \ 3.44(dd, \ J=6.2, \ 8.6 \ Hz, \ 1H), \ 7.40(d, \ J=8.6 \ Hz, \ 2H), \\ 1.91(m, \ 1H), \ 2.55(s, \ 3H), \ 3.32(s, \ 3H), \ 3.44(dd, \ J=6.2, \ 8.6 \ Hz, \ 1H), \ 7.40(d, \ J=8.6 \ Hz, \ 2H), \\ 1.91(m, \ 1H), \ 2.55(s, \ 3H), \ 3.32(s, \ 3H), \ 3.44(dd, \ J=6.2, \ 8.6 \ Hz, \ 1H), \ 7.40(d, \ J=8.6 \ Hz, \ 2H), \\ 1.91(m, \ 1H), \ 1.91(m, \ 1H), \$

7.73(d, J=8.6 Hz, 2H), 7.90-8.01(m, 5H), 10.48 (s, 1H).

Process 4

5

10

To a solution of 405 mg (0.85 mmol) of compound (XXII-1) in 3 ml of dichloromethane was added 3.3 ml (50 x 0.85 mmol) of trifluoroacetic acid and resulting mixture was stirred for 2 h at room temperature. The reaction mixture was concentrated in vacuo and the resulting residue was washed with ethyl ether to give 340 mg of the desired compound (Ia-4-1). Yield 94.7 %. mp. 231-234°C

IR(KBr, v max cm⁻¹): 1748, 1655, 1592, 1323, 1161.

Elemental analysis C19H22N2O5S2 · 0.1CF3COOH

Calcd. : C; 53.14 H; 5.13 N; 6.46 S; 14.78

Found: C; 53.48 H; 5.31 N; 6.57 S; 15.06

Example 177 - 208

The compounds which were shown in Tables 33 to 36 were synthesized in a manner similar to those described in Example 176.

Table 33

Γ	1	Т	·	T	T	T			
	Elemental analysis	l	C ₂₅ H ₂₃ N ₃ O ₆ S·0.9H ₂ O Calc. C:58.91 H:4.90 N:8.24 S:6.29 Foun.C:58.97 H:5.07 N:7.95 S:6.10	-	C ₂₄ H ₂₀ N ₄ O ₇ S•1.1H ₂ O Calc. C:54.56 H:4.24 N:10.60 S:6.07 Foun.C:54.51 H:4.32 N:10.83 S:6.15	C ₂₆ H ₂₆ N ₄ O ₅ S•0.9H ₂ O Calc. C:59.73 H:5.36 N:10.72 S:8.13 Foun.C:59.58 H:5.23 N:10.85 S:6.47	C ₂₅ H ₂₃ N ₃ O ₅ S•0.9H ₂ O Calc. C:60.82 H:5.06 N:8.51 S:6.49 Foun.C:60.83 H:5.19 N:8.66 S:6.66	C ₂₄ H ₂₀ BrN ₃ O ₅ S-0.6H ₂ O Calc. C:52.11 H:3.86 Br:14.44 N:7.60 S:5.80 Foun.C:52.13 H:4.04 Br:14.57 N:7.43 S:5.70	C ₂₅ H ₂₃ N ₃ O ₅ S ₂ ·0.7H ₂ O Calc. C:57.50 H:4.71 N:8.05 S:12.28 Foun.C:57.63 H:4.79 N:8.00 S:12.08
<i>(</i> -	IR (\(\nu\) cm ⁻¹) (KBr)	1732,1641 1341,1163	1726,1655 1323,1177	1723,1633 1361,1149	1719,1629 1340,1156	1732,1653 1399,1199	1731,1656 1591,1327 1160	1727,1668 1590,1316 1154	1728,1653 1593,1323 1159
H. 30214 + 0001 (m)	mp (decomp.) (C)	215-217	233-234	216-218	211-213	236-238	240-244	215-218	244-249
וואופר	*	æ	R	24	æ	24	~	8 2	æ
No. Y	R 18	-\\\-\\\-\\\-\\\\-\\\\\\\\\\\\\\\\\\\\	H3CO-C-N-C-N-CO26H	H-S-(-)-N2H	-N-3-(-)-N ² 0	(H3C)2N-Q-N-Q-N-Q-N-Q-N-Q-N-Q-N-Q-N-Q-N-Q-N-Q	H3C-{\\-_\-_\-_\-_\-_\-_\-_\-_\-_\	Br-G-N-G-N	H ₃ CS _ _ _ _ _ _ _ _ _ _ _ _ _
	. R	AT CHE	LY CH	N CH2.	IZ TO	IZ HO	IX HO	IZ HO	H CH ₂ -
	Example No.	177	178	179	180	181	182	183	184

	Elemental analysis	C ₂₁ H ₁₈ ClN ₃ O ₅ S Calc. C:54.84 H:3.94 Cl:7.71 N:9.14 S:6.97 Foun.C:54.39 H:4.06 Cl:7.42 N:8.98 S:6.99	C ₂₂ H ₂₀ ClN ₃ O ₅ S•0.1CF ₃ COOH Calc. C:55.15 H:4.19 Cl:7.33 N:8.69 S:6.63 Foun.C:55.25 H:4.28 Cl:7.10 N:8.80 S:6.80	C ₂₄ H ₂₄ N ₂ O ₅ S-0.5H ₂ O Calc. C:62.46 H:5.46 N:6.07 S:6.95 Foun.C:62.42 H:5.54 N:6.26 S:6.97	C ₁₉ H ₂₂ N ₂ O ₅ S-0.2H ₂ O Calc. C:57.91 H:5.73 N:7.11 S:8.14 Foun.C:57.94 H:5.69 N:7.03 S:8.14	C ₁₉ H ₂₂ N ₂ O ₅ S ₂ •0.1CF ₃ COOH Calc. C:53.14 H:5.13 N:6.46 S:14.78 Foun.C:53.48 H:5.31 N:6.57 S:15.06	C ₁₈ H ₁₉ FN ₂ O ₅ S•0.1CF ₃ COOH Calc. C:53.86 H:4.74 F:6.09 N:6.90 S:7.90 Foun.C:53.82 H:4.85 F:5.60 N:6.93 S:7.78	C ₁₈ H ₂₀ N ₂ O ₅ S•0.1H ₂ O Calc. C:57.16 H:5.38 N:7.41 S:8.48 Foun.C:57.01 H:5.46 N:7.57 S:8.57	C ₁₉ H ₂₂ N ₂ O ₆ S•0.2H ₂ O Calc. C:55.65 H:5.51 N:6.83 S:7.82 Foun.C:55.63 H:5.48 N:7.03 S:7.75
(1	IR (v cm·¹) (KBr)	1724,1673 1592,1326 1156	1725,1682 1592,1332 1160	1748,1659 1590,1324 1161	1749,1658 1592,1323 1161	1748,1655 1592,1323 1161	1749,1726 1668,1597 1322,1160	1728,1661 1591,1317 1159	1696,1654 1591,1317 1255
R¹8-SO₂NH ★COOH (Ia)	mp (decomp.) (C)	201-203	206-208	254-256	227-229	231-234	235-236	226-227	220-221
N-SONH N	*	æ	R	æ	æ	8	22	24	24
R ¹⁸ SC	8.18	CI—N—0—N—10	H ₃ C-N-C-N-C-N-C-N-C-N-C-N-C-N-C-N-C-N-C-N	-N-3-()	H ₃ C-\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	H ³ CS \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	F-0-N-0	N-0-0-	- H-0-C-W-00°H
	. R	CH2-CH2-	CH2-	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-
	Example No.	193	194	195	196	197	198	199	200

Example R_1 R_1 R_2 R_3 R_4 R_5 $R_$		mp (decomp.) (°C) 240-242 229-230 214-216	IR (v cm ⁻¹) (KBr) 1726,1688 1591,1347 1166 1726,1663 1592,1318 1159 1159 1159	Elemental analysis C ₁₈ H ₁₉ N ₃ O ₇ S-0.4H ₂ O Calc. C:50.44 H:4.66 N:9.80 S:7.48 Foun.C:50.40 H:4.55 N:9.90 S:7.44 C ₁₈ H ₁₉ BrN ₂ O ₅ S-0.2Ethylether Calc. C:48.03 H:4.50 Br:17.00 N:5.96 S:6.82 Foun.C:48.04 H:4.61 Br:16.83 N:5.96 S:6.86 C ₂₀ H ₂₄ N ₂ O ₆ S-0.4H ₂ O Calc. C:56.17 H:5.84 N:6.55 S:7.50 Foun.C:56.21 H:6.02 N:6.50 S:7.33
1 (CH ₃) ₂ CH- O ₂ N-\\ CH ₃ CH- O ₂ N-\\ CH ₃) ₂ CH- Br-\\ CH ₃ CO-\\ CH ₃) ₃ C- H ₃ CO-\\ \\ CH ₃ CO-\\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\	R R	229-230	1726,1688 1591,1347 1166 1726,1663 1592,1318 1159 1159 1318,1159	C ₁₈ H ₁₉ N ₃ O ₇ S-0.4H ₂ O Calc. C:50.44 H:4.66 N:9.80 S:7.48 Foun.C:50.40 H:4.55 N:9.90 S:7.44 C ₁₈ H ₁₉ BrN ₂ O ₅ S-0.2Ethylether Calc. C:48.03 H:4.50 Br:17.00 N:5.96 S:6.82 Foun.C:48.04 H:4.61 Br:16.83 N:5.96 S:6.86 C ₂₀ H ₂₄ N ₂ O ₆ S-0.4H ₂ O Calc. C:56.17 H:5.84 N:6.55 S:7.50 Foun.C:56.21 H:6.02 N:6.50 S:7.33
(CH ₃) ₂ CH· Br \ \times \cdot \cdo	R	229-230	1726, 1663 1592, 1318 1159 1659, 1591 1316, 1159	C ₁₈ H ₁₉ BrN ₂ O ₅ S-0.2Ethylether Calc. C:48.03 H:4.50 Br:17.00 N:5.96 S:6.82 Foun.C:48.04 H:4.61 Br:16.83 N:5.96 S:6.86 C ₂₀ H ₂₄ N ₂ O ₆ S-0.4H ₂ O Calc. C:56.17 H:5.84 N:6.55 S:7.50 Foun.C:56.21 H:6.02 N:6.50 S:7.33
(CH ₉) ₃ C- H ₉ CO $-\bigcirc$ C-N- \bigcirc R 214-216 \bigcirc CH ₂ - H ₃ C $-\bigcirc$ N- \bigcirc C-N- \bigcirc R 236-237 \bigcirc CH ₂ - N $-\bigcirc$ CH ₂ - N $-\bigcirc$ CH ₂ - N $-\bigcirc$ R 272-275		214-216	1659,1591 1316,1159	C ₂₀ H ₂₄ N ₂ O ₆ S-0.4H ₂ O Calc. C:56.17 H:5.84 N:6.55 S:7.50 Foun.C:56.21 H:6.02 N:6.50 S:7.33
5 CH2- CH2- N>C-N- R 272-275	R	236-237	1723,1679 1590,1337 1162	C ₂₁ H ₂₀ N ₄ O ₅ S·0.25CF ₃ COOH Calc. C:55.06 H:4.35 N:11.95 S:6.84 Foun.C:54.80 H:4.90 N:12.16 S:7.10
	R	272-275	1719,1672 1594,1339 1165	C ₂₁ H ₁₉ N ₃ O ₅ S Calc. C:59.28 H:4.50 N:9.88 S:7.54 Foun.C:58.84 H:4.56 N:9.71 S:7.36
$\frac{206}{11}$ $\frac{\text{H}_3\text{C}}{\text{N}_0}$ $\frac{\text{G-N}}{\text{G-M}}$ $\frac{17}{\text{R}}$ $\frac{17}{11}$	R	214-215	1733,1685 1594,1319 1154	C ₂₀ H ₁₉ N ₃ O ₆ S Calc. C:55.94 H:4.46 N:9.78 S:7.47 Foun.C:55.50 H:4.47 N:9.74 S:7.31
2 0 7		217-220	1732,1679 1592,1312 1155	-
2 0 8		-	1	1

Example 209 (Method E)

HCI
$$H_2N$$
 COO¹Bu $Process 1$ SO_2-N COO¹Bu $XXIII-1$

OHC SO_2-N COO¹Bu $Process 3$ $XXIII-1$

OHC SO_2-N COO¹Bu $Process 3$ $XXIV-1$

MeS SO_2-N COO¹Bu $Process 4$ $XXV-1$

MeS SO_2-N COO¹Bu S

Process 1

5

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To a solution of 20.94 g (99.8 mmol) of D-valine tert-butyl ester hydrochloride (XV-3) in 200 ml of dichloromethane was added 22 ml (2 x 99.8 mmol) of N-methylmorpholine and 20.27 g (99.8 mmol) of p-styrenesulfonyl chloride under ice-cooling. After being stirred for 15 h at room temperature, the reaction mixture was washed with 2N HCl, 5% NaHCO₃, water. The organic layer was dried over Na₂SO₄ and concentrated in vacuo, and the resulting residue was column chromatographed on silica gel. The fractions eluting with ethyl acetate / n-hexane / chloroform = 1/3/1 were collected and washed with n-hexane to give 28.93 g of the desired compound (XXIII-1). Yield 85 %. mp. 118-120℃.

 $IR(KBr, v max cm^{-1}): 3419, 3283, 1716, 1348, 1168.$

NMR(CDCl₃, δ ppm): 0.85(d, J=6.9 Hz, 3H), 1.00(d, J=6.6 Hz, 3H), 1.21(s, 9H), 2.04(m, 1H), 3.62(dd, J=9.8, 4.5 Hz, 1H), 5.09(d, J=9.8 Hz, 1H), 5.41(dd, J=0.5, 10.9 Hz, 1H), 5.84(dd, J=0.5, 17.6 Hz, 1H), 6.72(dd, J=10.9, 17.6 Hz, 1H), 7.49(d, J=8.4 Hz, 2H), 7.79(d, J=8.4 Hz, 2H).

Process 2

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Ozone gas was bubbled through a solution of 5.09 g (15 mmol) of compound (XXIII-1) in 300 ml of dichloromethane for 15 h at -78°C. To this solution was added 22 ml (20 x 15 mmol) of methylsulfide, and the reaction mixture was allowed to warm to room temperature gradually over 80 min and concentrated in vacuo to give 6.03g aldehyde derivative (XXIV-1).

IR(CHCl₃, v max cm⁻¹): 3322, 1710, 1351, 1170.

NMR(CDCl₃, δ ppm): 0.85(d, J=6.9 Hz, 3H), 1.00(d, J=6.9 Hz, 3H), 1.22(s, 9H), 2.07(m, 1H), 3.69(dd, J=4.5, 9.9 Hz, 1H), 8.01(s, 4H), 10.08(s, 1H).

15 Process 3

To a solution of 6.02 g(15 mmol) of compound (XXIV-1) in 60 ml of ethanol and 15 ml of tetrahydrofuran was added 2.72 g (1.05 x 15 mmol) of benzenesulfonyl hydrazide at room temperature. After being stirred for 2 h, the resulting mixture was concentrated in vacuo. The residue which was obtained by concentration in vacuo was column chromatographed on silica gel and the fractions eluting with chloroform / ethyl acetate = 1/4 were collected and recrystallized from ethyl acetate to give 4.44 g of the desired compound (XXV-1). Yield from process 2 60%. mp. 163-164°C.

Elemental analysis C22H29N3O6S2

Calcd. : C; 53.32 H; 5.90 N; 8.48 S; 12.94

Found: C; 53.15 H; 5.87 N; 8.32 S; 12.82

 $[\alpha]_D$ -11.6 ± 1.0(c=0.509 DMSO 23.5°C)

 $IR(KBr, v \max cm^{-1}): 3430, 3274, 1711, 1364, 1343, 1172.$

NMR(CDCl₃ δ ppm): 0.84(d, J=6.9 Hz, 3H), 0.99(d, J=6.6 Hz, 3H), 1.19(s, 9H), 2.00(m, 1H), 3.63(dd, J=4.5, 9.9 Hz, 1H), 5.16(d, J=9.9 Hz, 1H), 7.50-7.68(m, 5H), 7.73(s, 1H),

7.78-7.84(m, 2H), 7.96-8.02(m, 2H), 8.16(brs, 1H).

Process 4

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To a solution of 0.14 ml (1.11 x 1 mmol) of 4-(methylmercapto)aniline and 0.3 ml of conc. hydrochloric acid in 3 ml of aqueous 50% ethanol solution was added a solution of 78.4 mg (1.14 x 1 mmol) of sodium nitrite in 1 ml of water at 0 to 5 °C of the internal temperature and the reaction mixture was stirred for 15 min at the same temperature. To a solution of 496 mg (1 mmol) of compound (XXV-1) in 5 ml of dry pyridine was added the above reaction mixture over 8 min at -25°C. This reaction mixture was stirred for additional 4 h at -15°C to rt, poured into water, and extracted with ethyl acetate. The organic layer was washed with 2N HCl, 5% NaHCO₃, and water, dried over Na₂SO₄, and concentrated in vacuo. The residue was column chromatographed on silica gel and the fractions eluting with chloroform / ethyl acetate = 1/9 were collected to give 374 mg of the desired compound (XXVI-1). Yield 74%.

Elemental analysis C23H29N5O4S2 · 0.3H2O

Calcd. : C; 54.27 H; 5.86 N; 13.76 S; 12.60

Found: C; 54.25 H; 5.77 N; 13.87 S; 12.52

 $IR(KBr, v \max cm^{-1}): 3422, 3310, 1705, 1345, 1171.$

NMR(d₆-DMSO, δ ppm) : 0.83(d, J=6.9 Hz, 3H), 0.86(d, J=7.2 Hz, 3H), 1.19(s, 9H), 2.00(m, 1H), 2.59(s, 3H), 3.54(dd, J=6.3, 9.6 Hz, 1H), 7.56(d, J=8.7 Hz, 2H), 8.00(d, J=8.6 Hz, 2H), 8.10(d, J=8.7 Hz, 2H), 8.33(d, J=9.6 Hz, 2H), 8.34(d, J=8.7 Hz, 2H).

Process 5

A solution of 353 mg of compound (XXVI-1) in 2.5 ml of dichloromethane and 2.5 ml of trifluoroacetic acid was stirred for 3 h at room temperature. The reaction mixture was concentrated in vacuo and the resulting residue was washed with ethyl ether to give 308 mg of compound (Ia-5-1). Yield 98%. mp. 194 - 195°C.

 $IR(KBr, v max cm^{-1}): 1720, 1343, 1166.$

Elemental analysis C₁₉H₂₁N₅O₄S₂ · 1.1H₂O

Calcd. : C; 48.83 H; 5.00 N; 14.99 S; 13.72

Found: C; 49.13 H; 5.25 N; 14.55 S; 13.34

Example 210 - 251

The compounds which were shown in Tables 37 to 43 were synthesized in a manner similar to those described in Example 209.

Table 37

	¹ H-NMR(δ ppm) d ₆ -DMSO	_	2.65(dd,J=9.3,13.1Hz,1H),2.82(dd, J=5.8,13.1Hz,1H),3.86(dt,J=5.8,9.3 Hz,1H),7.72(A ₂ B ₂ q,J=8.1Hz,2H), 8.19(A ₂ B ₂ q,J=8.1Hz,2H),8.49(d,J= 9.3Hz,1H),8.88(s,1H),10.69(s,1H)
R¹®SO₂NH,*CONHOH (Ib)	IR (v cm·¹) (KBr)	ì	3700-2200(br),3278, 1634,1337,1160
R¹ O ₂ NH * CON	mp (decomp.)	ı	194-195
R ¹⁸ -S	*	×	Я
	R 18	R R	N=N N.N.
	. R	IZ J	-Z-CH2-
	Example No.	2 1 0	211

Table 38

	'H-NMR(8 ppm) ds-DMSO	1	2.75(dd,J=9.3,13.7Hz,1H),2.99(dd,J=5.3,13.7Hz,1H),3.96(dt,J= 5.3,9.3Hz,1H),8.53(d,J=9.3Hz, 1H)
	WN-H1		2.75(dd,J=9.3, dd,J=5.3,13.7 ¹ 5.3,9.3Hz,1H), 1H)
ООН (Іа)	IR (\(\nu\) cm ⁻¹) (KBr)	ţ	2400-3700br,3422,3337, 1733,1698,1347,1170
р¹ Р¹®SO₂NH + СООН (Ia)	mp (decomp.)	ļ	215-216
R. 8	*	æ	~
	R 18	N=N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N	N=N.N.
	R 1	N CH2-	-cH ₂ -
	Example No.	210	211

		÷		-R- 	-	
		No. 1	LIN2	R' SU2NH + CUUH		
Example No.	- <u>-</u>	R ! 8	*	mp (decomp.) (C)	IR (\(\nu\) cm ⁻¹) (KBr)	Elemental analysis
2 1 2	CH3 CH3 CH3	N=N.N.N.N.N.N.N.N.N.N.N.N.N.N.N.N.N.N.N	RS	199-202	1734,1337 1161	C ₂₅ H ₂₂ N ₆ O ₄ S·0.5Ethylether Calc. C:60.10 H:5.04 N:15.57 S:5.94 Foun.C:60.41 H:4.69 N:15.52 S:5.57
213	N N N N N N N N N N N N N N N N N N N	N=N N-N-N	RS	224-225	1728,1338 1166	C ₂₄ H ₁₉ FN ₆ O ₄ S•0.4Ethylether Calc. C:57.35 H:4.32 F:3.54 N:15.67 S:5.98 Foun.C:56.74 H:4.37 F:3.47 N:15.17 S:568
214	(CH ₃) ₂ CHCH ₂ -	N=Z.Y	Ж	202-204	1720,1595 1338,1170	C ₁₉ H ₂₁ N ₅ O ₄ S Calc. C:54.93 H:5.09 N:16.86 S:7.72 Foun.C:54.75 H:5.14 N:16.81 S:7.55
215	(СН3)2СН-	N=Z.N-	23	221-222	1696,1594 1349,1173	C ₁₈ H ₁₉ N ₅ O ₄ S Calc. C:53.38 H:4.83 N:17.29 S:7.92 Foun.C:53.38 H:4.80 N:17.05 S:7.67
216	***	N=N.N.N.N.N.N.N.N.N.N.N.N.N.N.N.N.N.N.N	RS	145-148	1727,1337 1163	I
217	-ZH2-CH2-	N=N.N.N.N.N.N.N.N.N.N.N.N.N.N.N.N.N.N.N	æ	203-205	1735,1495 1336,1160	C ₂₈ H ₂₃ N ₅ O ₄ S•0.6H ₂ O Calc. C:62.70 H:4.55 N:13.06 S:5.98 Foun.C:62.61 H:4.50 N:13.29 S:5.87
2 1 8		N=N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N	RS	225-227	1721,1418 1344,1163	C ₂₆ H ₂₁ N ₅ O ₄ S•0.2H ₂ O Calc. C:62.07 H:4.29 N:13.92 S:6.37 Foun.C:61.93 H:4.30 N:14.01 S:6.43
2 1 9	CHO CHO CH2-	N=N.N-N.N-N.N-N.N-N.N-N.N-N.N-N.N-N.N-N.	24	111-114	1727,1703 1459,1332 1165	C ₂₅ H ₂₀ N ₆ O ₅ S•H ₂ O Calc. C:56.17 H:4.15 N:15.72 S:6.00 Foun.C:56.20 H:4.18 N:15.68 S:6.10

Table 40

_									
	Elemental analysis	C ₂₅ H ₂₂ N ₆ O ₅ S Calc. C:57.91 H:4.28 N:16.21 S:6.18 Foun.C:57.77 H:4.29 N:16.01 S:6.37	C ₁₉ H ₂₁ N ₅ O ₄ S Calc. C:54.93 H:5.09 N:16.86 S:7.72 Foun.C:54.71 H:5.09 N:16.70 S:7.56	C ₂₀ H ₂₃ N ₅ O ₅ S-0.4H ₂ O Calc. C:53.06 H:5.30 N:15.47 S:7.08 Foun.C:53.13 H:5.13 N:15.12 S:7.14	-	C ₂₀ H ₂₃ N ₅ O ₅ S+0.4H ₂ O Calc. C:53.06 H:5.30 N:15.47 S:7.08 Foun.C:53.13 H:5.13 N:15.12 S:7.14	C ₁₆ H ₁₈ BrN ₅ O ₄ S-0.8H ₂ O Calc. C:43.70 H:3.99 Br:16.15 N:14.16 S:6.48 Foun.C:43.93 H:3.85 Br:15.92 N:13.87 S:6.47	1	
	IR (\(\nu\) cm ⁻¹) (KBr)	1749,1719 1331,1165	1730,1693 1349,1173	1729,1693 1337,1170	1718,1601 1385,1162	1719,1304 1162	1696,1348 1171	1698,1344 1168	1757,1738 1331,1163
H- 502NH + 1000 - H	mp (decomp.) (C)	195-196	205-207	204-207	190 decomp.	195-197	227-228	204-207	203-205
721417	*	R	R	В	R	24	R	R	24
٠ ٢	R <u>.</u>	H ₃ CO-(N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N	N=N.N.	H ₃ CO N=N	HO N=N,N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-	H ₃ CO N=N	Br N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-	H_3CO \longrightarrow $N=N$	F—N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-
	- -	N N N N N N N N N N N N N N N N N N N	СН3СН2(СН3)СН-	СН3СН2(СН3)СН-	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-	-Э ^ε (ЕНЭ)	H CH2.
	Example No.	220	221	222	223	224	225	226	227

Table 41

		•				
Example No.	. X	R 18	*	mp (decomp.) (C)	IR (v cm ⁻¹) (KBr)	Elemental analysis
228	СН ₂ -	Br N=N	R	197-199	1744,1325 1154	l
2 2 9	-CH2-	F ₃ C \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	æ	197-198	1738,1707 1328,1169	C ₂₃ H ₁₈ F ₃ N ₅ O ₄ S Calc. C:53.38 H:3.51 F:11.01 N:13.53 S:6.20 Foun.C:53.11 H:3.55 F:10.89 N:13.66 S:6.31
230	CH2-	O_2N	~	190-191	1730,1597 1345,1161	C ₂₂ H ₁₈ N ₆ O ₆ S•0.4H ₂ O Calc. C:52.67 H:3.78 N:16.73 S:6.39 Foun.C:52.73 H:3.92 N:16.53 S:6.55
2 3 1	-zho(F-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N	84	205-207	1730,1509 1236,1165	C ₂₂ H ₁₈ FN ₅ O ₄ S-0.2H ₂ O Calc. C:56.09 H:3.94 F:4.03 N:14.87 S:6.81 Foun.C:56.10 H:4.09 F:4.12 N:14.84 S:7.08
232	-2HDCH2-	CI—N=N,N—ID	~	204-206	1730,1493 1346,1164	C ₂₂ H ₁₈ CiN ₅ O ₄ S•0.6H ₂ O Calc. C:53.41 H:3.91 Ci:7.17 N:14.16 S:6.48 Foun.C:53.33 H:3.90 Ci:7.22 N:14.19 S:6.68
233	-ZHO-CH2-	H ₃ C \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	24	226-227	1732,1697 1509,1373 1345,1170	C ₂₃ H ₂₁ N ₅ O ₄ S•1.2H ₂ O Calc. C:56.94 H:4.86 N:14,44 S:6.61 Foun.C:56.88 H:4.49 N:14.31 S:6.72
234	-cH2CH2-	H ₃ CO V.N.N.	æ	214-216	1732,1697 1345,1168	C ₂₃ H ₂₁ N ₅ O ₅ S+1.7H ₂ O Calc. C:54.15 H:4.82 N:13.73 S:6.29 Foun.C:54.05 H:4.35 N:13.60 S:6.77
235	-cH2CH2-	N=N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N	24	190-192	1731,1605 1336,1160	C ₂₃ H ₁₈ N ₆ O ₄ S-0.8H ₂ O Calc. C:56.50 H:4.04 N:17.19 S:6.56 Foun.C:56.52 H:4.16 N:17.00 S:6.52

Table 42

R 1 R 18 \star mp (decomp.) IR (ν cm·1) Elemental analysis (C) (KBr)	Foun.C:61.59 H:5.45 N:13.85 S:6.34 Foun.C:61.59 H:5.45 N:13.89 S:6.27	C28H29N ₅ O ₄ S-0.3H ₂ O R 225-227 1739,1512 C3lc. C:62.62 H:5.56 N:13.04 S:5.97 Foun. C:62.46 H:5.52 N:13.43 S:6.28		→ CH ₂ - HO—	H N=N N=N R 205-207 1744,1716 C24H19BrN6O4S-1.7H2O C36. C:48.20 H:3.78 Br:13.36 N:14.05 S:5.36 Foun.C:48.27 H:3.75 Br:13.16 N:14.11 S:5.38	H N H ₃ C C ₂₅ H ₂₂ N ₆ O ₄ S-0.6H ₂ O R 199-201 1718,1685 Calc. C:58.49 H:4.56 N:16.71 S:5.90 Foun.C:58.52 H:4.69 N:16.71 S:5.90	H ₃ CH- H ₃ C	H ₃) ₂ CH- F————————————————————————————————————
R 1	CH2-	CH2-	-cH2-	-CH ₂ -	/=	/=	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-
Example No.	236	237	238	2 3 9	240	241	242	243

Table 43

٢		1			T		 -T		
	Elemental analysis	1	C ₁₉ H ₂₁ N ₅ O ₄ S ₂ *1.1H ₂ O Calc. C:48.83 H:5.00 N:14.99 S:13.72 Foun.C:49.13 H:5.25 N:14.55 S:13.34	C ₂₃ H ₂₁ N ₅ O ₄ S ₂ •0.2H ₂ O Calc. C:55.34 H:4.32 N:14.03 S:12.85 Foun.C:55.37 H:4.35 N:14.00 S:12.86	C ₂₅ H ₂₂ N ₆ O ₄ S ₂ •1.1H ₂ O Calc. C:54.16 H:4.40 N:15.16 S:11.57 Foun.C:54.20 H:4.66 N:15.09 S:11.62	C ₁₈ H ₁₆ N ₆ O ₄ S-0.4H ₂ O Calc. C:51.52 H:4.04 N:20.03 S:7.64 Foun.C:51.34 H:3.96 N:19.76 S:8.02		l	l
(la)	IR (\(\nu\) cm ⁻¹) (KBr)	1696,1348 1171	1720,1343 1166	1753,1497 1325,1165	1718,1677 1495,1333 1170	1698,1430 1327,1163	ļ	Î	ŀ
R ¹⁸ ·SO ₂ NH´ * COOH (I	mp (decomp.)	223-225	194-195	222-224	213-216	>220		1	-
NZ NH	*	R	В	ж	22	R	R	Ж	R
R ¹⁸⁻ S(R 18		H ₃ CS-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N	H ₃ CS-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N	H ₃ CS-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N	N=N HN,NH	H_2N $N=N$ $N=N$ $N=N$	$- \bigvee_{N=N}^{N-N} \bigvee_{N} - \bigvee_{N=N}^{N-N} SH$	OHC N=N N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N
		(СН₃)₂СН-	(CH ₃) ₂ CH-	CH ₂ -	IZ OH	N N N N N N N N N N N N N N N N N N N	-cH2-	-ZH2CH2-	CH ₂ -
	Example No.	244	245	246	247	248	249	250	251

Example 252 - 266

The compounds which were shown in Tables 44 to 45 were synthesized in a manner similar to those described in Example 157.

Table 44

_								· · · · · · · · · · · · · · · · · · ·	
	¹ H-NMR(δ ppm) d ₆ -DMSO	0.96(d,J=6.6Hz,3H) 1.01(d,6.8Hz,3H) 2.87(s,3H) 4.17(d,J=10.4Hz,1H)	0.71(d,J=6.6Hz,3H) 0.88(d,6.4Hz,3H) 2.88(s,3H) 3.48(d,J=10.8Hz,1H)	0.55(d,J=6.8Hz,3H) 0.82(d,6.6Hz,3H) 3.74(s,3H)	0.91(d,J=5.6Hz,6H) 1.52-1.69(m,4H) 3.84(d,J=10.4Hz,1H)	0.95(d,J=6.6Hz,3H) 0.97(d,6.8Hz,3H) 2.89(s,3H) 4.20(d,J=10.6Hz,1H)	0.92(d,J=6.6Hz,3H) 0.97(d,6.6Hz,3H) 2.84(s,3H) 4.73(t,J=7.4Hz,1H)	2.78(d.d,J=13.8,7.2Hz,1H) 3.14(d.d,J=14.8,7.4Hz,1H) 4.43(d,J=16.4Hz,1H) 4.68(d,J=16.4Hz,1H)	0.96(d,J=6.4Hz,3H) 0.97(d,J=6.4Hz,3H) 2.52(s,3H),2.93(s,3H)
	IR (\(\nu\) cm ⁻¹) (KBr)	1715,1583 1340,1151	3323,1678 1328,1150	3344,1684 1323,1149	3700-2200br 1681,1319 1212	3300-2400br 1711,1336 1185	3300-2400br 1719,1340 1153	3640-2400br 1736,1717 1694,1348 1162	3284br,1745 1714,1323 1131
	mp (decomp.) (C)	1	110-111	148-150	l	206-207	132-132.5	l	141-144
	*	R	24	R	æ	æ	æ	æ	R
P.3	R 20	нооэ-	-соинон	-соинон	нооэ-	-соон	-соон	Н000-	ноор-
	R 19	·cH³	-CH ₃	CH₂-	-(CH ₂) ₄ NH ₂	-CH3	-ÇH ²	CH2-	°HЭ-
	R 18	⟨ ⟩•-⟨⟩		₹ 0-0-{	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	N=N N-N	N=N.N.	N=N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N	H ₃ CS - Soft
	R1	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-	(CH ₃) ₂ CH-	(CH ₃) ₂ CHCH ₂ -	CH2-	(CH ₃) ₂ CH-
	Example No.	252	253	254	255	256	257	258	259

Table 45

	¹ H-NMR(δ ppm) ds-DMSO	0.72(d,J=6.4Hz,3H)0.85(d,J=6.4Hz,3H)2.47(s,3),4.15(d,J=10.2Hz,1H)4.51(d,J=15.5)Hz,1H)4.73(d,J=15.5Hz,1H)	2.54(s,3H),2.78(s,3H) 2.85(d.d,J=14.0,9.4Hz,1H) 3.16(d.d,J=14.0,6.0Hz,1H) 4.76(d.d,J=10.0,5.8Hz,1H)	1	1	ļ	1	1
	IR (\(\nu\) cm ⁻¹) (KBr)	3600-2400br 1718,1344 1151	3600-2400br 1719,1655 1592,1320 1154	j	ŧ	1	I	I
	mp (decomp.) (C)	1	_	-	ş	1	į	ŀ
E	*	ಜ	R	R	R	R	R	R
H ¹⁹	R 20	Н000-	НООЭ-	нооэ-	Н000-	нооэ-	Н000-	Н000-
	R 19	CH ₂ -	-CH3	⟨у-сн₂-	-(CH ₂) ₄ NH ₂	-CH3	CH2-	-(CH ₂)4NH ₂
	R 18	H ₃ CS-{}	-сн ₂ - М-8	-CH ₂ - H ₃ cs-{\rightarrow}-g-\frac{\rightarrow}{\rightarrow}-g-\rightarrow\frac{\rightarrow}{\rightarrow}-g-\frac{\rightarrow}{\rightarrow}-g-\rightarrow\rightarro	CH2- H3CO- CEC- S	-CH₂- H₃co-{_}-C≡c-{_}-	-CH ₂ - H ₃ CO-{}C≡C-{}-	-CH₂- H₃CO-{>-C≡C-{}-
	R 1	(СН₃)₂СН-	CH2-	CH2-	CH ₂ -	CH2-	CH2-CH2-	CH2-
	Example No.	260	261	262	263	264	265	266

Example 267

The compounds which were shown in Tables 46 were synthesized in a manner similar to those described in Example 92.

Table 46

٢	1		\neg
	'H-NMR(δ ppm) d ₆ -DMSO	1H-NMR(& ppm) de-DMSO 2.62(dd,J=8.4,13.5Hz,1H), 2.80(dd, J=6.0,13.5Hz,1H),3.82(ddd,J=6.0, 8.4,8.7Hz,1H),8.38(d,J=8.7Hz,1H)	
(IR (v cm·¹) (KBr)	3700-2400br,3267, 2217,1671,1321,1161	2200-3700br,3430, 3292,1728,1324,1162
E)	mp (decomp.)	156-158	176-178
•	*	Ж	æ
	R 20	-CONHOH R	-соон
	⊼ e-	C)-C≅C	√ >c≡c √
	٦.	CH2-CH2-	CH ₂
	Example No.	267	267

10

15

20

Test examples on the compounds of the present invention are described below.

The test compounds are the ones described in the Examples and Tables.

Test example

5 (1) Isolation and purification of MMP-9 (92 kDa, gelatinase B)

Type IV collagenase (MMP-9) was purified according to the methods descrived in the following literature. Scott M. Wilhelm et al., J. Biol. Chem., 264, 17213-17221, (1989), SV40-transformed Human Lung Fibroblasts Secrete a 92-kDa Type IV Collagenase Which Is Identical to That Secreted by Normal Human Macrophages; Yasunori Okada et al., J. Biol. Chem., 267, 21712-21719, (1992), Matrix Metalloproteinase 9 (92-kDa Gelatinase / Type IV Collagenase) from HT 1080 Human Fibrosarcoma Cells; Robin V. Ward et al., Biochem. J., (1991) 278, 179-187, The purification of tissue inhibitor of metalloproteinase-2 from its 72 kDa progelatinase complex.

MMP-9 is secreted from human fibrosarcoma cell line ATCC HT 1080, into its culture medium when it is stimulated with 12-tetradecanoylphorbol-13-acetate (TPA). The production of MMP-9 in this culture was verified by the gelatin zymography as described in the following literature (Hidekazu Tanaka et al., (1993) Biochem. Biophys. Res. Commun., 190, 732-740, Molecular cloning and manifestation of mouse 105-kDa gelatinase cDNA). The condition medium of the stimulated HT 1080 was concentrated and was purified with gelatin-Sepharose 4B, concanavalin A-sepharose, and Sephacryl S-200. The purified pro-MMP-9 (92 kDa, gelatinase B) thus obtained gave a single positive band in the gelatin zymography. Subsequently, activated MMP-9 was obtained by treating the pro-MMP-9 with trypsin.

25 (2) Assay methods of type IV collagenase inhibitors

Collagenase assay was performed using the activated MMP-9 described above and the substrate supplied in the type IV collagenase activity kit (YAGAI, inc.), according to the manufacturer's protocol. The following 4 assays are performed per compound (inhibitor).

- (A) substrate (type IV collagenase), enzyme (MMP-9), inhibitor
- (B) substrate (type IV collagenase), inhibitor
- (C) substrate (type IV collagenase), enzyme (MMP-9)
- (D) substrate (type IV collagenase)

According to the manufacturer's protocol, fluorescent intensity was measured and percent inhibition was determined by the following equation.

Inhibition (%) =
$$\{1 - (A - B) / (C - D)\} \times 100$$

 IC_{50} is a concentration at which the percent inhibition reaches 50 %. The results are shown in Tables 47 to 54.

Table 47

				
Example No.	Compound No.	IC ₅₀ (μM)	Compound No.	IC ₅₀ (μM)
1	1a-1-1	0.24	1b-1-1	0.030
2	1a-1-2	2.6	1b-1-2	0.04
3	1a-1-3	0.18	1b-1-3	0.005
4	1a-1-4	2. 25		
5	1a-1-5	0.81	1b-1-5	0.041
6	1a-1-6	0.68	1b-1-6	0.034
7			1b-1-7	0.028
8	1a-1-8	2. 0	1b-1-8	2. 0
9			1b-1-9	0.41
1 0			1b-1-10	2. 1
1 1			1b-1-11	1. 7
1 2			1b-1-12	0.085
1 3			1b-1-13	0.38
1 4	1a-1-14	3. 7	1b-1-14	0.11
1 5			1b-1-15	0.027
1 6	1a-1-16	0.520	1b-1-16	0.0108
1 7	1a-1-17	0.205	1b-1-17	0.0203
1 8	1a-1-18	0.500	1b-1-18	0.0282
2 0			1b-1-20	0.134
2 1	1a-1-21	4.65	1b-1-21	0.0041
2 3			1b-1-23	0.073
2 4			1b-1-24	0.2
2 6			1b-1-26	1. 3
2 7			1b-1-27	3. 0
3 0	la-1-30	1. 16	1b-1-30	0.213
3 1			1b-1-31	0.0129

Table 48

	 			
Example No.	Compound No.	IC ₅₀ (μ M)	Compound No.	IC ₅₀ (μM)
3 3	1a-1-33	0.24	1b-1-33	0.005
3 5	1a-1-35	2. 6	1b-1-35	0.0216
3 8	1a-1-38	0.018		
4 0	1a-1-40	0.076		
4 1	1a-1-41	0.312		
4 2	la-1-42	0.0123		
4 3	1a-1-43	0.625		
4 4	1a-1-44	1. 910		
4 5	la-1-45	0.040		
4 6	1a-1-46	1. 12		
4 7	1a-1-47	0.389		
4 8	1a-1-48	1. 15		
4 9	1a-1-49	0.249		
5 0	1a-1-50	0.553		
5 1	1a-1-51	0.110		
5 2	1a-1-52	0.329		
5 3	1a-1-53	1. 8		
5 4	1a-1-54	0.075		
5 5	1a-1-55	0.0398		
6 0	1a-1-60	1. 31	1b-1-60	0.0012
6 1	la-1-61	0.247	1b-1-61	0.247
6 2			1b-1-62	3. 50
6 3	1a-1-63	1. 05	1b-1-63	0.00039
6 4	1a-1-64	1. 90	1b-1-64	0.0037
6 5	1a-1-65	0.291	1b-1-65	0.0035

Table 49

Example No.	Compound No.	IC ₅₀ (μM)	Compound No.	IC ₅₀ (μ M)
6 7	1a-1-67		1b-1-67	0.0061
6 8	1a-1-68	0.231		
8 0	1a-1-80	1. 91		
8 3	1a-1-83	1.77		
8 5	1a-1-85	1. 2	1b-1-85	0.013
8 6	1a-1-86	0.35	1b-1-86	0.0053
8 7			1b-1-87	0.940
9 3	1a-2-2	0.237		
9 4	1a-2-3	0.0109		
9 5	1a-2-4	0.0759		
9 6	1a-2-5	0.123		
9 7	1a-2-6	0.088		
9 8	1a-2-7	0.0699		
100	1a-2-9	0.0577		
101	1a-2-10	0.023		
102	1a-2-11	0.0475		
1 0 3	1a-2-12	0.0981		
104	1a-2-13	3. 28		
1 0 5	la-2-14	2. 98		
106	1a-2-15	0.133		
1 0 7	1a-2-16	0.325		
1 0 9	1a-2-18	1. 19		<u> </u>
110	1a-2-19	0.203		
1 1 1	1a-2-20	3. 41		
1 1 2	1a-2-21	3. 74		
114	1a-2-23	0.929		<u> </u>

Table 50

Example No.	Compound No.	IC ₅₀ (μM)
1 1 5	la-2-24	0.161
1 1 7	1a-2-26	1. 19
1 1 8	1a-2-27	0.088
1 1 9	1a-2-28	1. 11
1 2 0	1a-2-29	1. 53
1 2 1	1a-2-30	0.0736
1 2 2	1a-2-31	0. 224
1 2 3	1a-2-32	0.0234
1 2 4	1a-2-33	0.0218
1 2 5	1a-2-34	0.0144
1 2 6	1a-2-35	0.156
1 2 7	1a-2-36	0.0243
1 2 8	1a-2-37	0.0922
1 2 9	1a-2-38	0.222
160	1a-3-2	0.040
161	1a-3-3	0.0108
162	1a-3-4	0.873
1 6 3	1a-3-5	0.0126
164	1a-3-6	0.0965
165	1a-3-7	0.230
166	1a-3-8	1. 28
1 6 7	1a-3-9	0.014
168	1a-3-10	0.0083
1 6 9	1a-3-11	0.244
1 7 0	1a-3-12	2.03
171	1a-3-13	0.0395

Table 51

	G 137	TC ()()
Example No.	Compound No.	IC ₅₀ (μM)
177	1a-4-2	0.684
178	1a-4-3	0.0252
179	1a-4-4	2.36
180	1a-4-5	0.045
181	1a-4-6_	0.0539
182	1a-4-7	0.0059
183	1a-4-8	0.0027
184	la-4-9	0.00325
185	1a-4-10	0.0422
186	la-4-11	0.0982
187	1a-4-12	0.177
188	1a-4-13	0.843
189	1a-4-14	0.0375
1 9 0	1a-4-15	0.0597
1 9 1	1a-4-16	0.0095
192	la-4-17	0.324
193	1a-4-18	0.722
1 9 5	1a-4-20	1. 1
196	1a-4-21	0.0573
197	1a-4-22	0.0161
198	1a-4-23	0.493
199	la-4-24	2.06
200	1a-4-25	0.173
2 0 1	1a-4-26	0.252
202	1a-4-27	0.0114
203	1a-4-28	0.173

Table 52

Example No.	Compound No.	IC ₅₀ (μM)	Compound No.	IC ₅₀ (μM)
204	1a-4-29	3. 95	-	
207	1a-4-30	4.44		
2 1 0	1a-5-2	0.024		
2 1 1	1a-5-3	0.210	1 b - 2 1 1	0.00565
2 1 2	1a-5-4	0.393		
2 1 3	1a-5-5	0.128		
2 1 4	1a-5-6	0.832		
2 1 5	1a-5-7	0.110		
2 1 6	1a-5-8	0.107		
2 1 8	1a-5-10	0.744		
2 1 9	1a-5-11	0.574		
2 2 0	1a-5-12	0.0167		
2 2 1	1a-5-13	0.316		
2 2 2	1a-5-14	0.078		
2 2 3	1a-5-15	0.349		
2 2 4	1a-1-16	0.0101		
2 2 5	la-5-17	0.0122		
2 2 6	1a-5-18	0.166		
2 2 7	1a-5-19	0.0198		
2 2 8	1a-5-20	0.106		
2 2 9	1a-5-21	0.215		
2 3 0	1a-5-22	0.281		
2 3 1	1a-5-23	0.197		
2 3 2	1a-5-24	0.144		
2 3 3	1a-5-25	0.0864		
2 3 4	1a-5-26	0.153		

Table 53

Example No.	Compound No.	IC ₅₀ (μM)	Compound No.	IC ₅₀ (μM)
2 3 5	1a-5-27	0.265		
2 3 6	1a-5-28	0.304		
2 3 7	1a-5-29	1. 32		
2 3 8	1a-5-30	2.85		
2 3 9	1a-5-31	0.243		
2 4 0	1a-5-32	0.0041		
2 4 1	1a-5-33	0.0131		
2 4 2	1a-5-34	0.0239		
2 4 3	1a-5-35	0.0529		
2 4 4	1a-5-36	0.0165		
2 4 5	1a-5-37	0.0059		
2 4 6	1a-5-38	0.0108		
2 4 7	1a-5-39	0.0035		
267	1a-2-66	1. 5	1b-2-66	0.011

Table 54

Example No.	Compound No.	IC ₅₀ (μM)
2 5 2	1-252	0.24
253	1-253	0.000039
254	1-254	0.00063
2 5 5	1-255	0.529
2 5 6	1-256	0.601
2 5 7	1-257	0.776
2 5 8	1-258	0.908
2 5 9	1-259	0.130
260	1-260	0.159
2 6 1	1-260	0.182

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The compound of the present invention showed strong activity for inhibiting type IV collagenase.

Industrial Applicability

It is considered that the compound of the present invention is useful to prevent or treat osteoarthritis, rheumatoid arthritis, corneal ulceration, periodontal disease, metastasis and invasion of tumor, advanced virus infection (e.g., HIV), arteriosclerosis obliterans, arteriosclerotic aneurysm, atherosclerosis, restenosis, sepsis, septic shock, coronary thrombosis, aberrant angiogenesis, scleritis, multiple sclerosis, open angle glaucoma, retinopathies, proliferative retinopathy, neovascular glaucoma, pterygium, keratitis, epidermolysis bullosa, psoriasis, diabetes, nephritis, neurodegengerative disease, gingivitis, tumor growth, tumor angiogenesis, ocular tumor, angiofibroma, hemangioma, fever, hemorrhage, coagulation, cachexia, anorexia, acute infection, shock, autoimmune disease, malaria, Crohn disease, meningitis, and gastric ulcer, because the compound of the present invention has strong inhibitory activity against metalloproteinase, especially MMP.

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CLAIMS

1. A composition for inhibiting metalloproteinase which contains a compound of the formula \underline{I} :

$$R^5 - R^4 - R^3 - SO_2 - N$$
 R^1
 R^2
 $R^3 - R^4 - R^3 - SO_2 - N$
 R^2

- wherein R^1 is optionally substituted lower alkyl, optionally substituted aryl, optionally substituted heteroaryl, or optionally substituted heteroarylalkyl; R^2 is hydrogen atom, optionally substituted lower alkyl, optionally substituted aryl, optionally substituted aryl, optionally substituted heteroarylalkyl; R^3 is a bond, optionally substituted arylene, or optionally substituted heteroarylene; R^4 is a bond, $\cdot(CH_2)m_{-}$, $\cdot CH=CH_{-}$, $\cdot C \equiv C_{-}$, $\cdot CO_{-}$, $\cdot CO_{-}NH_{-}$, $\cdot N=N_{-}$, $\cdot N(R^A)_{-}$, $\cdot NH_{-}CO_{-}NH_{-}$, $\cdot NH_{-}CO_{-}$, $\cdot O_{-}$, $\cdot S_{-}$, $\cdot SO_{2}NH_{-}$, $\cdot SO_{2}NH_{-}N=CH_{-}$, or tetrazol-diyl; R^5 is optionally substituted lower alkyl, optionally substituted $C_3 \cdot C_8$ cycloalkyl, optionally substituted aryl, optionally substituted heteroaryl, or an optionally substituted non-aromatic heterocyclic group; R^A is hydrogen atom or lower alkyl; Y is $\cdot NHOH$ or $\cdot OH$; and m is 1 or 2; provided $\cdot R^2$ is hydrogen atom when Y is $\cdot NHOH$, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.
- 2. A composition for inhibiting metalloproteinase which contains a compound of the formula \underline{I} :

$$R^5-R^4-R^3-SO_2-N$$
COY I

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wherein R¹ is optionally substituted lower alkyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaryl, or optionally substituted heteroarylalkyl; R² is hydrogen atom, optionally substituted lower alkyl, optionally substituted aryl, optionally substituted heteroaryl, or optionally substituted heteroarylalkyl; R³ is a bond, optionally substituted arylene, or

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optionally substituted heteroarylene; R4 is a bond, -(CH2)m-, -CH=CH-, -C = C-, -CO-, -CO-NH-, -N=N-, -N(RA)-, -NH-CO-NH-, -NH-CO-, -O-, -S-, -SO₂NH-, -SO₂-NH-N=CH-, or tetrazol-diyl; R5 is optionally substituted lower alkyl, optionally substituted C3-C8 cycloalkyl, optionally substituted aryl, optionally substituted heteroaryl, or an optionally substituted non-aromatic heterocyclic group; RA is hydrogen atom or lower alkyl; Y is -NHOH or -OH; and m is 1 or 2; provided R2 is hydrogen atom when Y is -NHOH, R5 is optionally substituted aryl or optionally substituted heteroaryl when R3 is optionally substituted arylene or optionally substituted heteroarylene and R4 is -CO-NH- or -NH-CO-, R5 is optionally substituted aryl or optionally substituted heteroaryl when R3 is optionally substituted arylene or optionally substituted heteroarylene and R4 is tetrazol-diyl, R5 is lower alkyl, aryl substituted by lower alkyl or optionally substituted aryl, or heteroaryl substituted by lower alkyl or optionally substituted aryl when R3 is optionally substituted arylene and R4 is a bond, both of R3 and R4 are not a bond at the same time, and R4 is not -O- when R3 is optionally substituted arylene or optionally substituted heteroarylene, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

- 3. A composition for inhibiting metalloproteinase of claim 1 or 2, which is a composition for inhibiting type-IV collagenase.
- 4. A compound of the formula \underline{I} :

$$R^5-R^4-R^3-SO_2-N$$
COY

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wherein R^1 is optionally substituted lower alkyl, optionally substituted aryl, optionally substituted heteroaryl, or optionally substituted heteroarylalkyl; R^2 is hydrogen atom, optionally substituted lower alkyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaryl, or optionally substituted heteroarylalkyl; R^3 is a bond, optionally substituted arylene, or optionally substituted heteroarylene; R^4 is a bond, $-(CH_2)m^2$, $-CH=CH^2$, $-C \equiv C^2$, $-CO^2$, -

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or tetrazol-diyl; R5 is optionally substituted lower alkyl, optionally substituted C3-C8 cycloalkyl, optionally substituted aryl, optionally substituted heteroaryl, or an optionally substituted non-aromatic heterocyclic group; RA is hydrogen atom or lower alkyl; Y is -NHOH or -OH; and m is 1 or 2; provided R² is hydrogen atom when Y is -NHOH, R5 is optionally substituted aryl or optionally substituted heteroaryl when R3 is optionally substituted arylene or optionally substituted heteroarylene and R4 is -CO-NH- or -NH-CO- (when R3 is phenylene and R4 is -CO-NH-, R1 is not methyl or phenyl and R⁵ is not 2-chlorophenyl, 4-chlorophenyl, or 2,4-dichlorophenyl), R⁵ is lower alkyl, optionally substituted aryl, or optionally substituted heteroaryl when R³ is optionally substituted arylene or optionally substituted heteroarylene and R4 is tetrazol-diyl, R5 is lower alkyl, aryl substituted with lower alkyl or optionally substituted aryl, or heteroaryl substituted with lower alkyl or optionally substituted aryl when R³ is optionally substituted arylene and R⁴ is a bond, both of R³ and R⁴ are not a bond at the same time, and R4 is not -O- when R3 is optionally substituted arylene or optionally substituted heteroarylene, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

5. A compound of the formula II:

$$R^7-R^6$$

$$SO_2-N$$

$$R^1$$

$$R^1$$

$$R^2$$

$$R^3$$

$$R^1$$

$$R^2$$

$$R^2$$

wherein R^6 is -CH=CH-, -C \equiv C-, -N=N-, -NH-CO-NH-, -S-, -SO₂NH-, or -SO₂-NH-N=CH-; R^7 is optionally substituted aryl or optionally substituted heteroaryl; R^8 and R^9 are each independently hydrogen atom, lower alkoxy, or nitro; R^1 , R^2 , and Y are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

6. A compound of the formula III:

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$$R^7 - R^{10} - SO_2 - N - COY$$

wherein R¹⁰ is -(CH₂)m-, -CO-, -CO-NH-, -N(R^A)-, -NHCO-, or tetrazol-diyl; m is 1 or 2; R¹, R², R⁷, R⁸, R⁹, R^A, and Y are as defined above, provided R¹ is not methyl or phenyl and R⁷ is not 2-chlorophenyl, 4-chlorophenyl, or 2,4-dichlorophenyl when R¹⁰ is -NH-CO-, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

7. A compound of the formula \underline{IV} :

$$R^7 - R^{11} \longrightarrow SO_2 - N \longrightarrow COY$$

wherein R^{11} is a bond, -CH=CH-, or -C \equiv C-; X is oxygen atom or sulfur atom; R^1 , R^2 , R^7 , and Y are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

8. A compound of the formula \underline{I} :

$$R^{5'}-R^{4'}-R^{3'}-SO_2-N$$
 $R^{2'}$
 $R^{2'}$
 $R^{2'}$

wherein R¹' is benzyl, (indol-3-yl)methyl, (1-methylindol-3-yl)methyl, (5-methylindol-3-yl)methyl, (5-fluoroindole-3-yl)methyl, (1-acetylindol-3-yl)methyl, (1-methylsulfonylindol-3-yl)methyl, (1-alkoxycarbonyl-3-yl)methyl such as ethoxycarbonylmethyl, or i-propyl; R²' is hydrogen atom, methyl, 4-aminobutyl, or benzyl; R³' is 1,4-phenylene; R⁴' is -O-; R⁵' is phenyl or 4-hydroxyphenyl; and Y is as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

9. A compound of the formula \underline{I} :

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wherein R1" is 4-thiazolylmethyl, (indol-3-yl)methyl, (5-methoxyindol-3-yl)methyl, 1naphthylmethyl, 2-naphthylmethyl, 4-biphenylylmethyl, 2,2,2-trifluoroethyl, 2phenylethyl, benzyl, i-propyl, 4-nitrobenzyl, 4-fluorobenzyl, cyclohexylmethyl, (1methylindol-3-yl)methyl, (5-methylindol-3-yl)methyl, (5-fluoroindol-3-yl)methyl, (pyridin-4-yl)methyl, (benzothiazol-2-yl)methyl, (phenyl)(hydroxy)methyl, phenyl, 2-carboxyethyl, hydroxymethyl, phenylmethoxymethyl, carboxymethyl, carboxybenzyl, (benzimidazol-2-yl)methyl, (1-methylsulfonylindol-3-yl)methyl, or (1ethoxycarbonylindol-3-yl)methyl; R2" is hydrogen atom; R3" is 1,4-phenylene; R4" is a bond; R5" is phenyl, 3-methoxyphenyl, 4-methoxyphenyl, 4-methylphenyl, 4-tert-4-trifluoromethylphenyl, 4-fluorophenyl, 4-methylthiophenyl, butvlphenyl. biphenylyl, 2-thienyl, benzoxazol-2-yl, benzothiazol-2-yl, or tetrazol-2-yl; and Y is as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

10. A compound of the formula $\underline{\mathbf{V}}$:

$$R^{7}-R^{12}$$
 $SO_{2}-N$ $COOH$ Y

wherein R^{12} is -CH=CH- or -C \equiv C-; R^1 , R^2 , R^7 , R^8 , and R^9 are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

1.1. A compound of the formula $\underline{\mathbf{VI}}$:

wherein R², R⁸, and R⁹ are as defined above, R¹³ is optionally substituted lower alkyl,

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optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaryl, or optionally substituted heteroarylalkyl; and R¹⁴ is optionally substituted aryl or optionally substituted heteroaryl; provided R¹³ is not methyl or phenyl and R¹⁴ is not 2-chlorophenyl, 4-chlorophenyl, or 2,4-dichlorophenyl, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

1 2. A compound of the formula VII:

$$\begin{array}{c|c}
N & R^8 & R^1 \\
R^7 - N & SO_2 - N & COOH & VII \\
R^9 & R^9
\end{array}$$

wherein R¹, R², R⁷, R⁸, and R⁹ are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

13. A compound of the formula VIII:

wherein R¹, R², R⁷, and R¹¹ are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

14. A compound of the formula <u>IX</u>:

$$R^7-O$$

$$= SO_2-N$$

$$= R^9$$

$$= R^9$$

$$= R^9$$

$$= R^1$$

$$= R^0$$

$$= R^0$$

$$= R^0$$

wherein R¹, R², R⁷, R⁸, and R⁹ are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

1 5. A compound of the formula \underline{X} :

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$$R^7 - R^{12} - SO_2 - N + COOH X$$

wherein R^{12} is -CH=CH- or -C \equiv C-; R^1 , R^7 , R^8 , and R^9 are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

1 6. A compound of the formula XI:

$$R^{14}$$
- $C-N$
 R^{8}
 R^{13}
 R^{14} - $C-N$
 R^{9}
 R^{13}
 R^{13}
 R^{14}

wherein R¹, R⁸, R⁹, R¹³, and R¹⁴ are as defined above, provided R¹³ is not methyl or phenyl and R¹⁴ is not 2-chlorophenyl, 4-chlorophenyl, or 2,4-dichlorophenyl, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

17. A compound of the formula XII:

$$R^7 - N N SO_2 - N COOH XII$$

wherein R¹, R⁷, R⁸, and R⁹ are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

18. A compound of the formula XIII:

- wherein R¹, R⁷, and R¹¹ are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.
 - 1 9. A compound of the formula XIV:

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$$R^7-O$$

$$SO_2-N$$

$$R^9$$

$$COOH$$

$$XIV$$

wherein R¹, R⁷, R⁸, and R⁹ are as defined above, its optically active substance, their pharmaceutically acceptable salt, or hydrate thereof.

- 2 0. The compound of any one of claims 4 to 19, wherein R¹, R¹, R¹, and R¹³ are i-propyl, benzyl, or (indole-3-yl)methyl.
 - 2 1. The compound of any one of claims 4 to 7 and 10 to 19, wherein R⁵, R⁷, and R¹⁴ are phenyl optionally substituted with one or more substituents selected from the group consisting of alkoxy, alkylthio, and alkyl.
 - 2 2. The compound of any one of claims 4 to 19, wherein a configuration of asymmetric carbon atoms bonding with R¹, R¹, R¹, and R¹³ is R configuration.
 - 2 3. A pharmaceutical composition containing a compound of any one of claims 4 to 19.
 - 2 4. A composition for inhibiting metalloproteinase containing a compound of any one of claims 4 to 19.
- 15 2 5. A composition for inhibiting type IV collagenase containing a compound of any one of claims 4 to 19.

ABSTRACT

Compounds having a metalloproteinase inhibitory activity, represented by the formula (I), its optically active isomers, their pharmaceutically acceptable salts, or bydrates thereof.

$$R^{5}-R^{4}-R^{3}-SO_{2}-N$$
 R^{1}
 R^{2}
 R^{2}

DECLARATION AND POWER OF ATTORNEY

As a below named inventor, I hereby declare that:

My residence, post office address, and citizenship are as stated below next to my name.

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled **SULFONATED AMINO ACID DERIVATIVES AND METALLOPROTEINASE INHIBITORS CONTAINING THE SAME** the specification of which is attached hereto.

I hereby state that I have reviewed and understand the contents of the above-identified specification, including the claims, as amended by any amendment referred to above.

I acknowledge the duty to disclose information which is known by me to be material to patentability as defined in Title 37, Code of Federal Regulations § 1.56.

I hereby claim foreign priority benefits under Title 35, United States Code, § 119(a)-(d) or § 365(b) of any foreign application(s) for patent or inventor's certificate, or § 365(a) of any PCT International application which designated at least one country other than the United States, listed below and have also identified below any foreign application for patent or inventor's certificate, or PCT International application having a filing date before that of the application on which priority is claimed:

PRIOR FOREIGN APPLICATION(S)

NUMBER	COUNTRY DAY/MONTH/YEAR FILED		PRIORITY CLAIMED
8/30082	Japan	23/01/96	Yes
8/213555	Japan	13/08/96	Yes

hereby claim the benefit under Title 35, United States Code § 119(e) of any United States provisional application(s) listed below.

	APPLICATION NO.	FILING DATE
and condition		

hereby claim the benefit under Title 35, United States Code, § 120 of any United States application(s), or § 365(c) of any PCT International application designating the United States, listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States or PCT International application in the manner provided by the first paragraph of Title 35, United States Code, § 12, I acknowledge the duty to disclose information which is known by me to be material to patentability as defined in Title 37, Code of Federal Regulations § 1.56 which became available between the filing date of the prior application and the national or PCT International filing date of this application:

APPLICATION SERIAL NO.	FILING DATE	STATUS: PATENTED, PENDING, ABANDONED
PCT/JP97/00126	22/01/97	Pending

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I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

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